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Discovery of benthol A and its challenging stereochemical assignment: opening up a new window for skeletal diversity of super-carbon-chain compounds†

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Super-carbon-chain compounds (SCCCs) are marine organic molecules featuring long polyol carbon chains with numerous stereocenters. Polyol–polyene compounds (PPCs) and ladder-frame polyethers (LFPs) are two major families. It is highly challenging to establish the absolute configurations of SCCC. In this century, few new SCCC families have been reported. Benthol A, an aberrant SCCC, was obtained from a South China Sea benthic dinoflagellate that should belong to a new taxon. Its planar structure and absolute configuration, containing thirty-five carbon stereocenters, were unambiguously established by a combination of extensive NMR spectroscopic investigations, periodate degradation of the 1,2-diol groups, ozonolysis of the carbon–carbon double bonds, *J*-based configurational analysis, NOE interactions, modified Mosher's MTPA ester method, and DFT-NMR ¹³C chemical-shift calculations aided by DP4+ statistical analysis. Benthol A displayed potent antimalarial activity against *Plasmodium falciparum* 3D7 parasites. This new molecule combines extraordinary structural features, particularly eight scattered ether rings on a C₇₂ backbone chain, which places it within a new SCCC family between PPCs and LFPs, herein termed polyol–polyether compounds. This suggestion was strongly supported by principal component analysis. The discovery of benthol A does not only provide new insights into the untapped biosynthetic potential of marine dinoflagellates, but also opens up a new window for skeletal diversity of SCCC.

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Introduction

Marine dinoflagellates produce massive and diverse super-carbon-chain compounds (SCCCs), the backbones of which

are long and flexible carbon chains that are highly oxygen-functionalized.^{1,2} Representative examples are palytoxin and maitotoxin with molecular weights of 2680 and 3422 daltons, respectively. To date, maitotoxin is believed to have the longest carbon chain in nature, except for biopolymers.^{1–4} SCCC exhibit remarkable bioactivities, such as activation effects of ion channels, and antifungal, antitumor, and anti-osteoclastic activities.^{1–8} However, the highly time-consuming stereochemical determination of SCCC retards their potential applications in biomedicine.

There is no doubt that fragment stereodivergent synthesis combined with NMR spectroscopic data comparison is an accurate and reliable approach to the stereochemical assignment of SCCC.^{1–8} However, it normally takes decades to determine the configuration of a complex SCCC. For example, symbiodinolide, a SCCC containing a 62-membered lactone and 61 carbon stereocenters, was first identified in 2007 from the symbiotic dinoflagellate, *Symbiodinium* sp.⁹ Since then, chemists have constantly striven to resolve its absolute configuration through fragment stereodivergent synthesis. Nevertheless, the absolute configurations of the last two fragments on its backbone, viz., C53–C61 and C75–C79, remain to be determined.^{2,10–12}

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With the aid of modern molecular biological, algaculture engineering, chemical, spectroscopic, and computational techniques, the configurational determination of complex SCCCs based on natural chemical entity has become feasible, though it still takes years. For instance, gibbosols A–C were obtained by our group in 2017 as a new class of SCCCs, *i.e.*, polyol-polyol compounds, from the marine pelagic dinoflagellate, *Amphidinium gibbosum*. The planar structures and absolute configurations of these SCCCs, containing thirty-six or thirty-seven carbon stereocenters, were successfully established in 2020 by combined use of chemical, spectroscopic, and computational approaches based on natural products.^{13,14}

During the ongoing projects in search of novel SCCCs from South China Sea dinoflagellates, we turned our attention to new taxa of these unicellular algae. Comparison of the 18S rDNA sequence of a benthic dinoflagellate (strain: MDRC-02, Fig. S1†) with those in NCBI nucleotide database using the BLAST algorithm revealed that the closest relative is an uncultured eukaryote clone (XM.18S_7; percent identity: 99.65%), whereas the closest dinoflagellate strain is *Amphidinium operculatum* (strain: TAK-0; percent identity: 80.95%). Phylogenetic tree inferred from the 18S rDNA sequence alignment in the NCBI databases suggested that the monophyly of the benthic strain MDRC-02 and the uncultured eukaryote clone XM.18S_7 formed a distinct clade, which is not closely related to any other clades of the dinoflagellate strains (Fig. S2†). Hence, the benthic dinoflagellate strain MDRC-02 should belong to a new taxon of dinoflagellate, or at least a new species of the genus *Amphidinium*. UPLC-MS-guided isolation led to the discovery of an unprecedented SCCC, named benthol A (**1**) (Fig. 1A), from the above benthic strain MDRC-02. Its chain backbone possesses eight scattered ether rings. Unquestionably, this raises the enormous opportunity to discover novel SCCCs from new taxa of dinoflagellates. Herein, we report the isolation, structural elucidation, and bioactivity of **1**. The principal component analysis score plot of **1** and previously reported 187 SCCCs strongly supports that **1** belongs to a new SCCC family.

Results and discussion

Planar structure of benthol A

Benthol A (**1**) was obtained as a colorless solid. The molecular formula $C_{75}H_{126}O_{30}$ with thirteen degrees of unsaturation was established by means of high resolution-electrospray ionization-mass spectrometry (HR-ESIMS) with a positive ion at m/z 1529.8234 ($[M + Na]^+$, calcd 1529.8226) (Fig. S7–S8†). According to the 1H and ^{13}C NMR spectroscopic data of **1** (Fig. S9–S22†), five degrees of unsaturation are due to five carbon-carbon double bonds. Thus, the molecule should contain eight rings.

The ^{13}C NMR spectroscopic data and the DEPT135 experiment of **1** (Fig. S23–S27†), combined with its 2D HSQC and sel-HSQC spectra (Fig. S28–S33†), revealed the presence of 75 carbon resonances for two methyl groups, 29 methylene groups (an olefinic, two oxygenated, and two overlapped signals at δ_C 41.73), 40 methine groups (six olefinic and 34 oxygenated, including two overlapped oxymethine signals at δ_C 73.22), and four quaternary

carbons (three olefinic). After fed with $NaH^{13}CO_3$ in the culture medium of the dinoflagellate, ^{13}C isotope enriched chemical entity of **1** (4.0 mg) was obtained for the 2D INADEQUATE experiment (Fig. S34–S36†).^{15,16} Consequently, three substructures, *viz.*, **I** (from C1 to C40, and C73), **II** (from C41 to C52, and C74), and **III** (from C53 to C72, and C75), were determined by analysis of 2D INADEQUATE, HSQC, H2BC, HMBC, and HSQC-TOCSY correlations of **1** (Fig. S28–S41†). It is worth mentioning that, most carbon-carbon connections of the molecule were established by 2D INADEQUATE correlations (Fig. 1B).

For substructure **I**, the linear connections of C1 to C18, C19 to C22, and C23 to C38 were revealed by the corresponding ^{13}C – ^{13}C correlations observed from the 2D INADEQUATE spectrum, and the relevant H2BC and HMBC crosspeaks. The linear connection of C18 and C19 and that of C22 and C23 were resolved by H2BC correlations between H18/C19 and H22/C23, respectively, whereas the branched connection of the C10=C73 bond was assigned by the INADEQUATE correlation between C10/C73. The linear connection of C38 to C40 was deduced from HSQC-TOCSY correlations between C39/H38 and C40/H39. Ether bridges between C4/C8, C8/C12, C21/C24, C29/C32, and C35/C39 were determined by crucial HMBC crosspeaks between H4/C8, H12/C8, H24/C21, H32/C29, and H39/C35, respectively. Taken together, the above data gave substructure **I**, containing three tetrahydropyran (*i.e.*, rings A, B, and E) and two tetrahydrofuran (*i.e.*, rings C and D) moieties, among which rings A and B (from C4 to C12) constitute a 6,6-spiroketal motif (Fig. 1B).

For substructure **II**, the linear connection of C41 to C51 was established by the corresponding INADEQUATE and H2BC correlations. The linear connection of C51 and C52 was revealed by the H2BC crosspeak between H52/C51 and the HSQC-TOCSY correlation between C51/H52, whereas the branched connection of Me-74 and C44 was assigned by the INADEQUATE correlation between C74/C44, the H2BC crosspeak between H₃-74/C44, and HMBC correlations between H₃-74/C43 and H₃-74/C45. An ether bridge between C49/C52 was determined by the crucial HMBC crosspeak between H49/C52. Taken together, the substructure **II**, containing a tetrahydrofuran moiety (*i.e.*, ring F), was elucidated (Fig. 1B).

For substructure **III**, the linear connections of C53 and C54, and C56 to C72 were deduced by the corresponding INADEQUATE and H2BC correlations. The linear connection of C54 to C56 was resolved by H2BC correlations between H55/C54 and H55/C56, and HSQC-TOCSY crosspeaks between C54/H55 and C56/H55, whereas the branched connection of Me-75 and C60 was assigned by the INADEQUATE correlation between C75/C60, the H2BC crosspeak between H₃-75/C60, and HMBC correlations between H₃-75/C59 and H₃-75/C61. Ether bridges between C54/C58 and C68/C72 were determined by crucial HMBC crosspeaks between H54/C58 and H₂-72/C68, respectively. Taken together, the substructure **III**, containing two tetrahydropyran moieties (*i.e.*, rings G and H), was assembled (Fig. 1B).

The presence of five tetrahydropyran and three tetrahydrofuran rings in **1**, including a 6,6-spiroketal motif, was further confirmed by isotope shift experiments measured in CD_3OD and CD_3OH .¹⁷ Fifteen oxygenated carbons, *viz.*, C8 (δ 99.76), C49 (δ



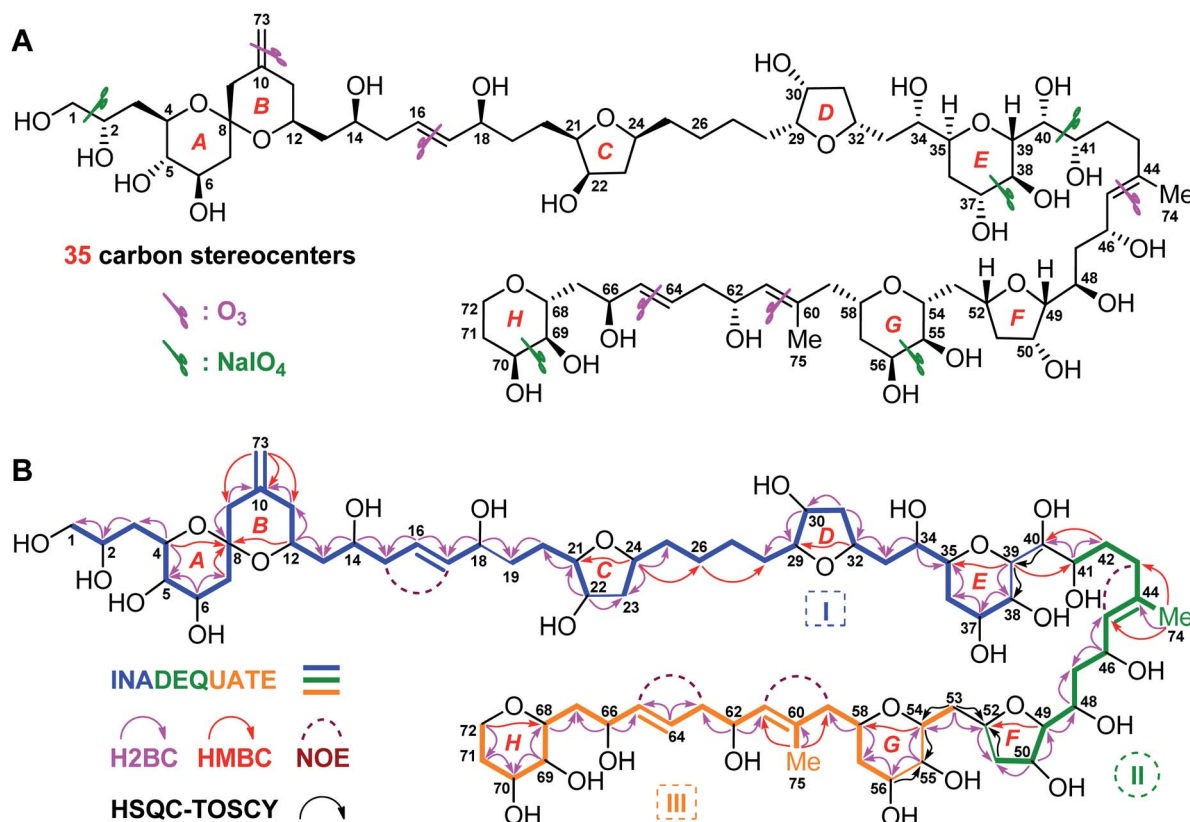


Fig. 1 (A) Structure and absolute configuration of benthol A (**1**). (B) Key INADEQUATE, H2BC, HMBC, and HSQC-TOCSY correlations and diagnostic NOE interactions of benthol A (**1**).

86.13), C29 (δ 84.69), C21 (δ 84.52), C24 (δ 79.06), C35 (δ 77.77), C52 (δ 76.54), C32 (δ 75.39), C68 (δ 75.01), C54 (δ 74.33), C4 (δ 73.54), C39 (δ 73.22), C58 (δ 70.96), C12 (δ 68.65), and C72 (δ 62.63), which exhibited no obvious deuterium-induced isotope shift, are involved in the ether linkages (Fig. S42–S46[†]). These results were in accordance with the aforementioned ether bridges assigned by the observed HMBC crosspeaks.

Finally, the H2BC correlation between H41/C40 and HMBC crosspeaks between H39/C41 and H₂-42/C40 led to the connection of substructures **I** and **II** through the C40–C41 bond. Similarly, H2BC correlations between H₂-53/C52 and the HSQC-TOCSY crosspeak between C53/H52 connected substructures **II** and **III** by the C52–C53 bond. NOE interactions between H₂-15/H17, H₂-43/H45, H₂-59/H61, and H₂-63/H65, and the large values of $^3J_{\text{H16,H17}}$ (15.4 Hz) and $^3J_{\text{H64,H65}}$ (15.4 Hz) concluded that the geometries of four carbon–carbon double bonds of **1**, *viz.*, C16=C17, C44=C45, C60=C61, and C64=C65, are all *E*-forms. Based on the above results, the planar structure of **1**, containing an exomethylene moiety, two pendant methyl groups, eight ether rings, and twenty-two hydroxy groups as 1,2-diol, 1,3-diol, 1,4-diol, and 1,5-diol moieties, was completely elucidated (Fig. 1B).

Chemical degradation of benthol A

Owing to heavily overlapped ^1H and ^{13}C NMR signals of **1**, particularly those of oxymethine groups crowded around

73.2 ppm, the determination of the relative configuration of **1** turned out to be a daunting task. To unambiguously establish the stereochemistry of the whole molecule, degradation reactions were applied to **1**. $\text{NaIO}_4/\text{NaBH}_4$ -mediated cleavage of the 1,2-diol groups of **1** with ^{13}C natural abundance afforded two main products, *viz.*, fragments **1A** (from C2 to C40, and Me-73) and **1B** (from C41 to C72, Me-74, and Me-75) (Fig. 2), whereas O_3/NaBH_4 -mediated cleavage of the carbon–carbon double bonds of **1** with ^{13}C -enriched abundance yielded five products, *viz.*, fragments **1a** (from C1 to C16), **1b** (from C17 to C44, and Me-74), **1c** (from C45 to C60, and Me-75), **1d** (from C61 to C64), and **1e** (from C65 to C72) (Fig. 2). The planar structures of these fragments were assigned by ESIMS, NMR spectroscopic data, and ^1H – ^1H COSY, HSQC, HMBC, and NOESY spectra (Fig. S47–S144[†]). Most notably, **1b** is a pair of epimers at C44.

Relative configuration of benthol A

To establish the relative configurations of the relevant segments in fragments **1A**, **1B**, **1a**, **1b**, **1c**, and **1e** (Fig. 2), extensive *J*-based configurational analysis (*J*BCA)^{18,19} and NOE interactions were employed. In fact, the relative configurations of four flexible segments, including C32–C35 in **1A** and C46–C49, C52–C54, and C66–C68 in **1B**, were assigned by *J*BCA referring to the typical criteria (Table S1[†]), whereas those of eight cyclic ether moieties (*i.e.*, rings A–H) in the above fragments were determined by diagnostic NOE interactions. To sum up, the relative

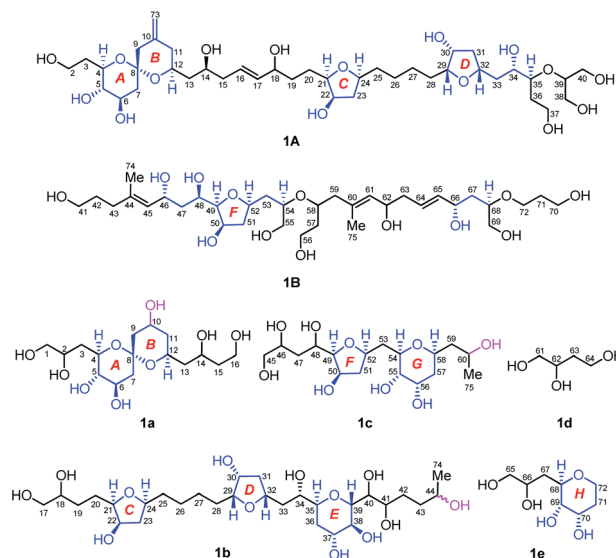


Fig. 2 Structures of seven degradation fragments, viz., **1A**, **1B**, and **1a–1e**. The relative configuration of each segment in blue color was assigned. Stereocenters at C10, C44, and C60 were generated as results of chemical degradation. Fragments **1a** and **1c** are optical isomers obtained from HPLC chiral separation, whereas **1b** is a mixture of epimers at C44.

configurations of the C32–C35 segment in **1A**, and the C46–C49, C52–C54, and C66–C68 segments in **1B** were assigned as (*anti/syn*), (*anti/anti*), *syn*, and *syn*, respectively, by application of *J*BCA (Fig. S3†).

The relative configuration of the 6,6-spiroketal moiety in **1a**, including rings A (from C4 to C8) and B (from C8 to C12), were assigned by diagnostic NOE interactions and the distinctive values of $^3J_{H,H}$ (Fig. 3A). The chair conformation of ring A with the axial H4, H5, and H6 was revealed by NOE interactions between H4/H6, H5/H3a, H5/H3b, and H5/H7 α , the large values of $^3J_{H4,H5}$ (9.3 to 9.6 Hz), $^3J_{H5,H6}$ (9.3 Hz), and $^3J_{H6,H7\alpha}$ (9.0 Hz), and the intermediate value of $^3J_{H6,H7\beta}$ (5.2 Hz). Similarly, the chair conformation of ring B with the axial H12 and the equatorial H10 was supported by NOE interactions between H12/H2, H12/H4, and H12/H14, the large values of $^3J_{H12,H11\beta}$ (9.8 Hz) and $^3J_{H12,H13\beta}$ (9.9 Hz), and the intermediate values of $^3J_{H10,H9\beta}$ (5.4 Hz) and $^3J_{H10,H11\beta}$ (5.4 Hz). Most notably, the relative configuration of C8 in **1a** was unequivocally assigned as *S*^{*} based on NOE interactions between H7 α /H9 α , H7 β /H9 β , H4/H14, H6/H14, and H4/H12 (Fig. 3A). Finally, the relative configuration of the 6,6-spiroketal moiety in **1a** was determined to be (4*R*^{*},5*S*^{*},6*R*^{*},8*S*^{*},12*S*^{*}).

NOE interactions between H29/H32 and H30/H32 of ring D in **1b** revealed their cofacial relationship (Fig. 3B). The chair conformation of ring E in **1b** with the axial H37, H38, and H39, and the equatorial H35 was corroborated by NOE interactions between H34/36 β , H34/H37, H34/H39, H37/H39, and H36 α /38, the large values of $^3J_{H37,H38}$ (9.6 Hz) and $^3J_{H38,H39}$ (9.6 Hz), and the small value of $^3J_{H35,H36\beta}$ (1.4 Hz) (Fig. 3B). Combined with the aforementioned (*anti/syn*) relationship of the C32–C35 segment in **1A** (Fig. S3A†), the relative configuration of the C29–C39 moiety in **1b** was concluded to be (29*R*^{*},30*R*^{*},32*R*^{*},34*S*^{*},35*S*^{*},37*R*^{*},38*S*^{*},39*R*^{*}).

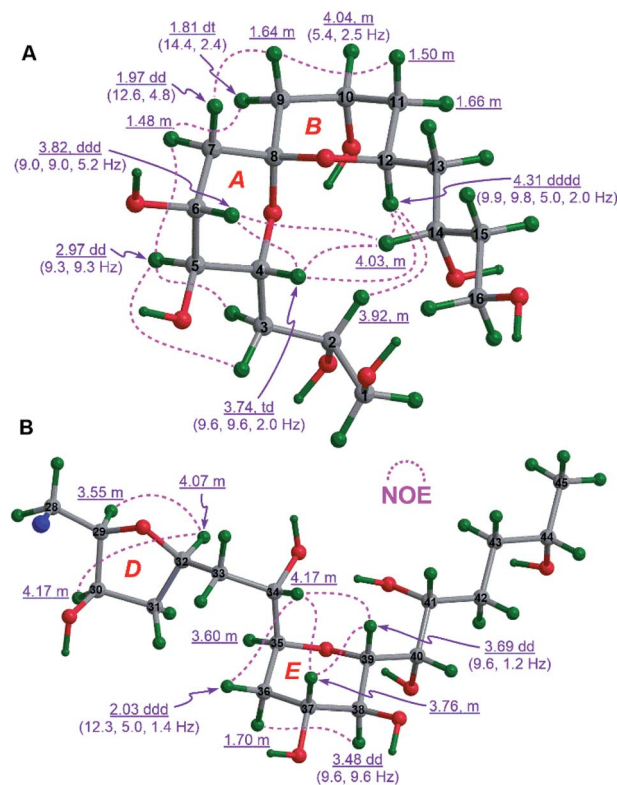


Fig. 3 (A) The relative configuration of the 6,6-spiroketal moiety in **1a** and (B) those of rings D and E in **1b** assigned by diagnostic NOE interactions and $^3J_{H,H}$ values (the C17 to C27 moiety in **1b** is omitted for clarity).

Additionally, the cofacial relationship of H21, H22, and H24 of ring C in **1b** was deduced from NOE interactions between H21/H24 and H22/H24 (Fig. S4A†).

The cofacial relationship of H49, H50, and H52 in ring F of **1c** was assigned by NOE interactions between H49/H52 and H50/H52, whereas the chair conformation of ring G in **1c** with the axial H54, H55, and H58, and the equatorial H56 was supported by those between H54/H58 and H55/H57 β , the large $^3J_{H54,H55}$ (9.6 to 9.8 Hz), and the intermediate $^3J_{H55,H56}$ (3.2 Hz) and $^3J_{H56,H57\beta}$ (5.8 Hz). Interestingly, the strong NOE interaction between H52/H54 revealed their same spatial orientation (Fig. 4A). Combined with the aforementioned (*anti/anti*) configuration of the C46–C49 segment (Fig. S3B†) and the *syn* relationship of the C52–C54 segment observed in **1B** (Fig. S3C†), the relative configuration of **1c** was thereby assigned as (46*R*^{*},48*R*^{*},49*R*^{*},50*R*^{*},52*R*^{*},54*R*^{*},55*S*^{*},56*S*^{*},58*S*^{*}).

Similarly, the chair conformation of ring H in **1e** with the axial H68 and H69, and the equatorial H70 was supported by NOE interactions between H68/H72 α and H69/H71 β , the large $^3J_{H68,H69}$ (9.6 Hz), and the intermediate $^3J_{H69,H70}$ (2.8 to 3.0 Hz) and $^3J_{H70,H71\beta}$ (5.4 to 5.8 Hz) (Fig. 4B).

Absolute configuration and bioactivities of benthol A

To determine the absolute configurations of **1A**, **1B**, **1a**, **1b**, **1c**, and **1e**, the modified Mosher's α -methoxy- α -trifluoromethyl-



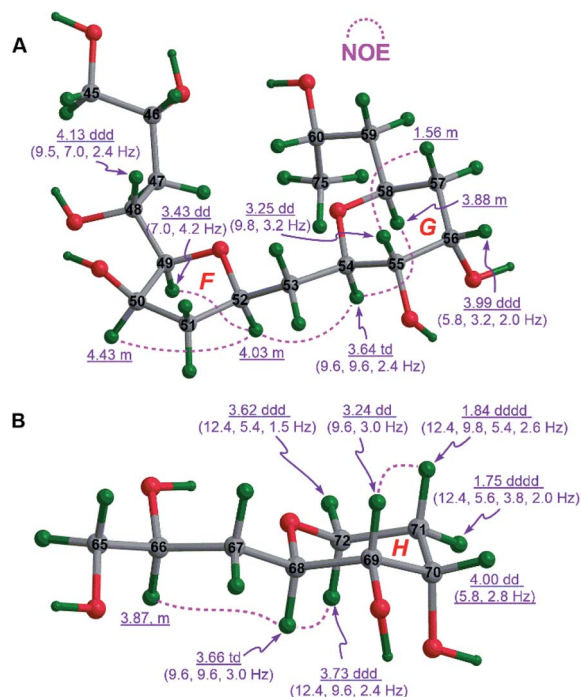


Fig. 4 (A) The relative configurations of rings F and G in **1c** and (B) that of ring H in **1e** assigned by diagnostic NOE interactions and $^3J_{H,H}$ values.

phenylacetic acid (MTPA) ester method was employed. The fragments **1A**, **1B**, **1b**, and **1c** were esterified with (*R*)- and (*S*)-MTPACl to produce the corresponding pairs of (*S*)- and (*R*)-MTPA esters (Fig. 5), viz., **1As/r** (Fig. S145–S158†), **1Bs/r** (Fig. S159–S172†), **1bs/r** (Fig. S185–S200†), and **1cs/r** (Fig. S201–S212†), respectively, whereas the fragments **1a** and **1e** reacted first with pivaloyl chloride to protect their terminal primary alcohol groups and then with (*R*)- and (*S*)-MTPACl to produce the corresponding pairs of (*S*)- and (*R*)-MTPA esters (Fig. 5), viz., **1as/r** (Fig. S173–S184†) and **1es/r** (Fig. S213–S222†), respectively.

Comparison of the $\Delta\delta^{SR}$ values between **1as/r**, the absolute configurations of C2, C6, C10, and C14 in **1a** were assigned as *S*, *R*, *R*, and *S*, respectively, by typical Mosher ester analysis²⁰ (Fig. 5). Based on the relative configuration of **1a** (Fig. 3A), the absolute configurations of C4, C5, C8, and C12 in **1a** were concluded to be *R*, *S*, *S*, and *R*, respectively (Fig. 5). Owing to the different CIP priority, the stereodescriptor at C5 in **1a** is opposite to that of **1as/r**. Similarly, the absolute configurations of C18, C22, and C41 in **1b** were determined to be *S*, *R*, and *S*, respectively, by typical Mosher ester analysis.²⁰ It is the same with **22R** in **1A**. Most notably, the above assigned **18S** and **22R** configurations were consistent with the $\Delta\delta^{SR}$ sign distribution proposed for the acyclic 1,5-diol systems, as noted by Riguera,^{21–27} viz., a type D *syn*-1,5-diol for the bis-MTPA esters of the C14–C18 diol moiety and a type B *anti*-1,5-diol for that of the C18–C22 diol moiety in **1A** (Fig. 5). Based on the relative configuration of ring C in **1b** (Fig. S4A†), the absolute configurations of C21 and C24 were concluded to be *R* and *S*, respectively. The absolute configurations of C34 and C37 in **1b** were

thereby assigned as *S* and *R*, respectively, by the sign distribution of $\Delta\delta^{SR}$ values proposed for the 1,4-diol systems, i.e., a type A *syn*-1,4-diol in the bis-MTPA esters of the C34–C37 diol moiety (Fig. 5).^{21–27} Based on the aforementioned (**29R***,**30R***,**32R***,**34S***,**35S***,**37R***,**38S***,**39R***) configuration of the C29–C39 segment in **1b** (Fig. 3B), the absolute configuration of this segment was successfully determined to be (**29R**,**30R**,**32R**,**34S**,**35S**,**37R**,**38S**,**39R**). Owing to the different CIP priority, the stereodescriptor at C38 in **1b** is opposite to that of **1bs/r**.

The absolute configurations of C46 and C48 in both **1B** and **1c** were identically assigned as *R* and *R*, respectively, by the sign distribution of $\Delta\delta^{SR}$ values proposed for the 1,3-diol systems, i.e., a type A *anti*-1,3-diol for the bis-MTPA esters of the C46–C48 diol moiety in both **1B** and **1c**.^{21–27} The absolute configuration of C50 in **1B** was established as *R* by typical Mosher ester analysis (Fig. 5).²⁰ Based on the relative configuration of ring F in **1c** (Fig. 4A), the absolute configurations of C49 and C52 were assigned as *R* and *R*, respectively. Due to the different CIP priorities, the stereodescriptors at C49 and C52 in **1c** are opposite to that of **1cs/r**. The absolute configurations of C55 and C56 in **1c** were assigned as *S* and *S*, respectively, by the sign distribution of $\Delta\delta^{SR}$ values proposed for the 1,2-diol systems, i.e., a type D *anti*-1,2-diol in the bis-MTPA esters of the C55–C56 diol moiety in **1c** (Fig. 5).^{21–27} Owing to the different CIP priority, the stereodescriptor at C55 in **1c** is opposite to that of **1cs/r**. Based on the relative configuration of ring G in **1c** (Fig. 4A), the absolute configurations of C54 and C58 were concluded to be *R* and *S*, respectively. The above results were consistent with the aforementioned relative configuration of **1c**.

The absolute configurations of C62 and C66 in **1B** were determined to be *R* and *S*, respectively, by the $\Delta\delta^{SR}$ sign distribution proposed for the acyclic 1,5-diol systems, i.e., a type A *anti*-1,5-diol for the bis-MTPA esters of the C62–C66 diol moiety in **1B** (Fig. 5).^{21–27} According to the *syn* relationship of the C66–C68 segment in **1B** (Fig. S3D†), the absolute configuration of C68 was thereby assigned as *R*. Based on the relative configuration of ring H in **1e** (Fig. 4B), the absolute configurations of C69 and C70 in **1e** were determined to be *S* and *S*, respectively (Fig. 5). Most notably, the **62R** configuration was reconfirmed based on the positive specific rotation ($[\alpha]_D^{25} = +8$ (*c* = 0.02, methanol)) of fragment **1d** (Fig. 2),^{28,29} whereas the **66S** configuration was further supported by the positive $\Delta\delta^{SR}$ sign values of H₂-65 and methyl protons of the terminal pivaloyl ester in **1e** (Fig. 5). In addition, these results were corroborated by a type D *anti*-1,2-diol in the bis-MTPA esters of the C69–C70 diol moiety in **1e**,^{21–27} though minor disturbance from the MTPA esters of C66 could be observed in the $\Delta\delta^{SR}$ value of H₆₉. Owing to the different CIP priority, the stereodescriptor at C69 in **1e** is opposite to that of **1es/r** (Fig. 5).

Thus far, the absolute configuration of C40 remained to be established. Due to its overlapped proton and carbon signals in **1b**, JBCA could not be applied. To tackle this problem, DFT-NMR ^{13}C chemical-shift calculations and DP4+ statistical analysis were employed.^{30,31} A model compound **1b'** representing the C34–C41 region of **1b** was chosen for the calculation. The Merck Molecular Force Field (MMFF) conformational search was



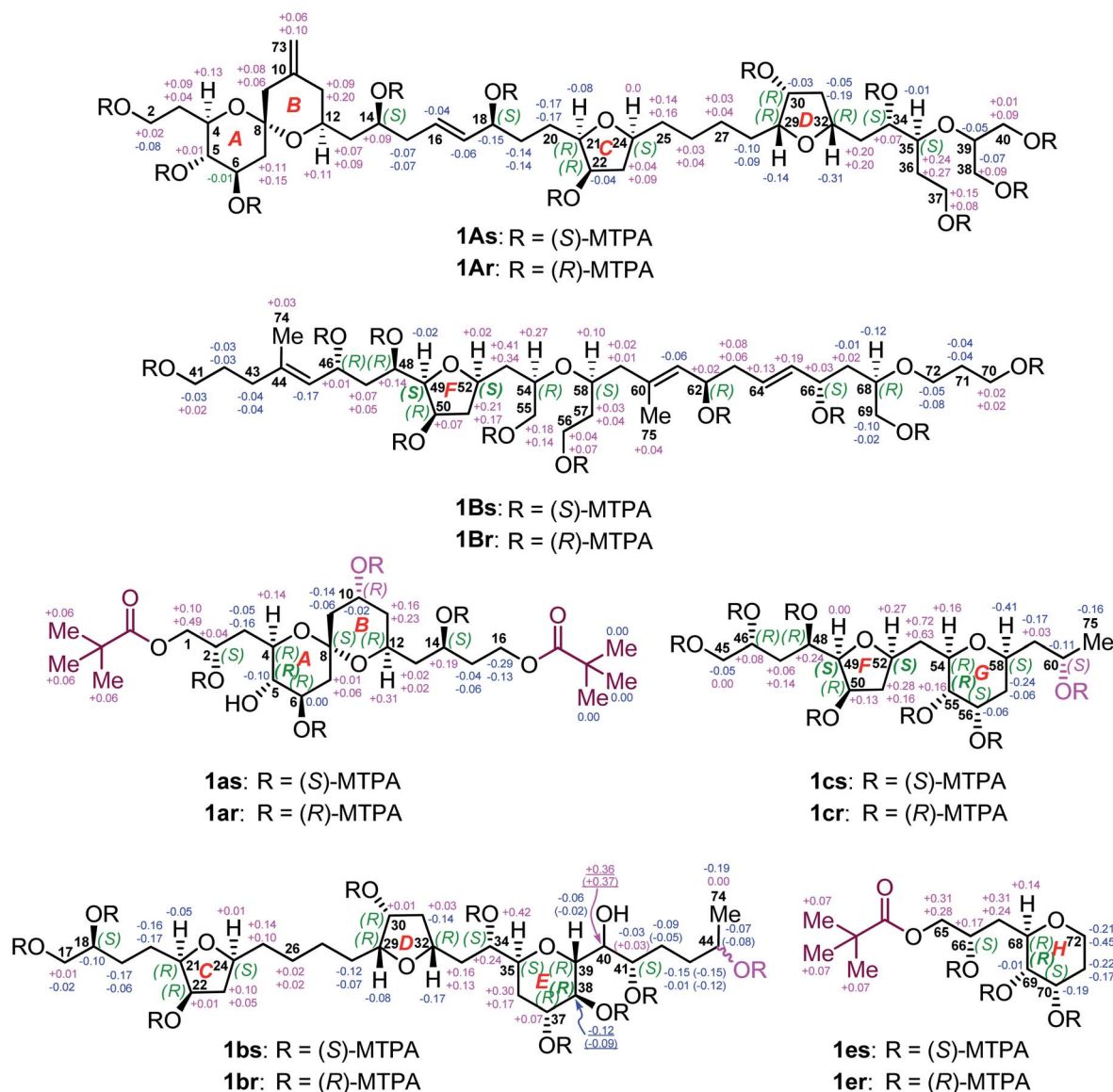


Fig. 5 $\Delta\delta^{SR}$ values obtained for fragments 1A, 1B, 1a, 1b, 1c, and 1e. Due to the different Cahn–Ingold–Prelog (CIP) priorities, the stereo-descriptors at C49 and C52 of 1B, C5 of 1a, C38 of 1b, C49, C52, and C55 of 1c, and C69 of 1e, are opposite to those of 1Bs/r, 1as/r, 1bs/r, 1cs/r, and 1es/r, respectively. Owing to the epimeric C44 in 1bs/r, $\Delta\delta^{SR}$ values of the C38–C44 segment and the Me-74 group are in pairs. However, 1as/r and 1cs/r are 10*R* and 60*S* isomers, respectively.

carried out on two possible C-40 epimers, viz., (34*S*,35*S*,37*R*,38*S*,39*R*,40*R*,41*S*)-1b' and (34*S*,35*S*,37*R*,38*S*,39*R*,40*S*,41*S*)-1b' (Fig. S4B†), producing 97 and 83 conformers, respectively, in a 21 kJ mol^{−1} energy window.^{32,33} The B3LYP/6-31+G(d,p) re-optimization of these conformers resulted in eight and twelve low-energy geometries with Boltzmann population above 1%. ¹³C NMR chemical-shift calculations were performed at the mPW1PW91/6-311+G(2d,p) level (Table S2†).³⁴ The carbon chemical-shift differences and the corrected mean average error value³⁵ showed a clear preference for the 40*R* epimer. The most characteristic chemical-shift difference could be observed at C41, adjacent to C40, with a $\Delta\delta$ value of 0.01 ppm for the 40*R* isomer *versus* that of 3.76 ppm for the 40*S* one (Table S3†). The DP4+ statistical

analysis resulted in 99.93% confidence for the 40*R* epimer. Additionally, 3.41 Hz was computed for the $^3J_{H39,H40}$ value of the 40*R* epimer, whereas 6.75 Hz was obtained for that of the 40*S* epimer. Evidently, the experimental $^3J_{H39,H40}$ (1.2 Hz) further supported the absolute configuration of 40*R* in 1b.

Considering all the assignments of the absolute configurations for thirty-five carbon stereocenters in seven degradation products, viz., 1A, 1B, and 1a–1e, we may conclude that the absolute configuration of benthol A (1) is (2*S*,4*R*,5*S*,6*R*,8*S*,12*S*,14*S*,18*S*,21*R*,22*R*,24*S*,29*R*,30*R*,32*R*,34*S*,35*S*,37*R*,38*S*,39*R*,40*R*,41*S*,46*R*,48*R*,49*R*,50*R*,52*R*,54*R*,55*S*,56*S*,58*S*,62*R*,66*S*,68*R*,69*S*,70*S*) (Fig. 1A).

Nowadays, malaria is still one of the most deadly infectious diseases in the world, particularly in Africa and Asia. Due to



rapid spread of resistant *Plasmodium* parasites, it is in urgent need to discover the next generation of potent antimalarial drugs with new chemical entities. Thus, benthol A (**1**) was assayed *in vitro* against the drug sensitive 3D7 strain of *P. falciparum*.³⁶ As a consequence, it exhibited antiplasmodial activity with an EC₅₀ value of 60.80 nM. Dihydroartemisinin was used as a positive control with an EC₅₀ value of 4.12 nM. In addition, *in vitro* anti-HIV-1 assay^{37,38} revealed that **1** showed an inhibition rate of 57.1% at the concentration of 100.0 μM and 16.5% at the concentration of 20.0 μM, respectively. Efavirenz was used as a positive control with 90.2% inhibition at the concentration of 100.0 μM and 88.5% at the concentration of 20.0 μM, respectively.

Principal component analysis

To date, 187 SCCCs with molecular weights of greater than 1000 daltons have been isolated from marine dinoflagellates of eight genera, viz., *Amphidinium*, *Gambierdiscus*, *Karenia*, *Karlodinium*, *Ostreopsis*, *Prorocentrum*, *Protoceratium*, and *Symbiodinium* (Table S4†). Among them, polyol-polyene compounds (PPCs) and ladder-frame polyethers (LFPs), including 60 and 90 members, respectively, are two major families and occupy 79.8% of the total SCCCs.

PPCs are mainly produced by dinoflagellates of the genera *Amphidinium*,^{39–63} *Karlodinium*,^{64–69} *Ostreopsis*,^{70,71} and *Prorocentrum*.^{72,73} These SCCCs are characterized by the presence of both a polyol and a polyene chain, connected by a central core containing two tetrahydropyran rings. Due to numerous stereocenters on flexible acyclic carbon chains of PPCs, only the absolute configurations of three molecules, i.e., AM3,^{61,63} KmTx2,^{65,67} and ostreol B,⁷¹ have hitherto been completely established. LFPs, featuring *trans*-fused polycyclic ether scaffolds, are generated by marine dinoflagellates of the genera *Gambierdiscus*,^{74–76} *Karenia*,^{17,77–79} and *Protoceratium*.^{80,81} Marine

toxins, such as ciguatoxins, brevetoxins, maitotoxin, and yes-sotoxins, are the representative examples.^{74–81} Owing to the universality of NOE interactions and ECD cotton effects for the stereochemical assignment of LFPs, the absolute configurations of these marine toxins have been determined.

Principal component analysis (PCA) is a widely used statistical technique for exploratory data analysis.⁸² It can reduce the dimensionality of complex data sets with the aim to achieve visualization loading in two- or three-dimensional plots.⁸³ To categorize the SCCCs reported to date, PCA was applied to 188 SCCCs, including **1** and previously explored 187 SCCCs (Fig. 6). Eight parameters were considered, including molecular weight, and carbon numbers, ring numbers, hydroxy numbers, carbon-carbon double bond numbers, chiral carbon numbers, and spiroketal carbon numbers of the backbone chain, and the ratio of backbone carbons to ether rings. The model obtained by PCA showed a good fitting degree ($R^2X = 0.996$) and a strong predictive ability [$Q^2(\text{cum}) = 0.882$] (Fig. S5†). The PCA score plot revealed that PPCs and LFPs are the two largest independent groups. Except for PPCs, the other SCCCs from dinoflagellates of the genus *Amphidinium* are dispersively distributed. Most notably, **1** is distinctly separated from all the other SCCCs. It cannot be grouped with any other SCCCs from dinoflagellates of the genus *Amphidinium* (Fig. 6). The nearest neighbors of **1** are dinophysistoxin-4 (DTX-4, **131**), DTX-5a (**132**), DTX-5b (**133**), DTX-5c (**134**), and prorocentric acid (**137**) (Fig. 6 and S6†), all of which were obtained from dinoflagellates of the genus *Prorocentrum*. The above four DTX analogs are the derivatives of okadaic acid, containing three spiroketal moieties within the C₃₈ backbone chain, whereas prorocentric acid comprises a 6,8-dioxabicyclo[3.2.1]octane motif and two tetrahydropyran rings within the C₆₀ backbone chain. The molecule of **1**, however, contains eight scattered ether rings within the C₇₂ backbone chain.

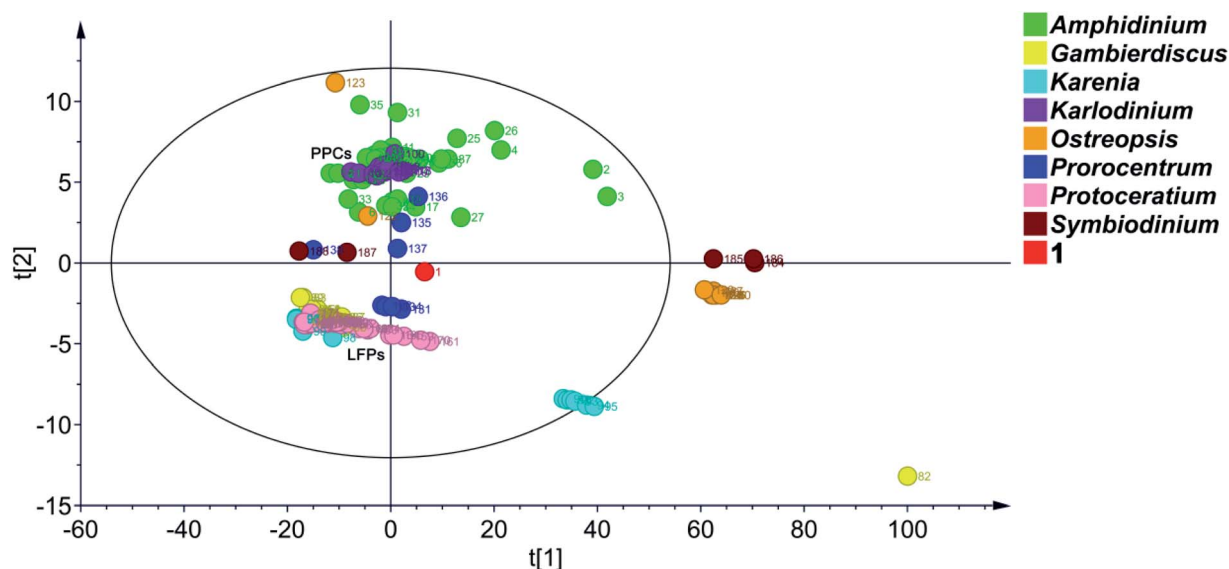


Fig. 6 PCA score plot of 188 SCCCs. Benthol A is designated as the number **1** (The red dot) and previously reported 187 SCCCs as numbers 2 to 188 in the graph. $R^2X = 0.996$, $Q^2(\text{cum}) = 0.882$, ellipse: Hotelling's T2 (95%).



Accordingly, **1** should be classified as the first member of a new SCCC family situated between PPCs and LFPs.

Obviously, SCCCs from dinoflagellates of three genera, *viz.*, *Amphidinium*, *Ostreopsis*, and *Prorocentrum*, are dispersively distributed, whereas those from the genera *Karenia* and *Symbiodinium* could be categorized into two main groups. SCCCs from the other three genera, *viz.*, *Karlodinium*, *Gambierdiscus*, and *Protoceratium*, are concentrated. Maitotoxin (**82**), however, is isolated, mainly due to its huge molecular weight (Fig. 6). It may be concluded that dinoflagellates of the genera *Amphidinium*, *Ostreopsis*, and *Prorocentrum* harbor SCCCs with the richest structural diversity.

In fact, the first member of most SCCC families was reported in the last century. As aforementioned, symbiodinolide (**184**) was identified with a 62-membered lactone in 2007.⁹ However, its analog zooxanthellatoxin-A (**186**) was reported in 1995.¹² The above two SCCCs are grouped with the inseparable isomeric mixture zooxanthellamides C₁–C₅ (**185**)⁸⁴ to form a distinct SCCC family, termed super macrolides (Fig. 6).

It is worth mentioning that, amdigenols A (**2**) and D (**3**), containing three tetrahydropyran rings within the C₉₈ backbone chain and four tetrahydropyran moieties within the C₁₀₁ backbone chain, respectively, were obtained as dimeric AM analogs from the Okinawan dinoflagellate *Amphidinium* sp. in 2012 and 2020, respectively. The two SCCCs are grouped to form an independent SCCC family in the PCA score plot (Fig. 6).

It is reported that PPCs, such as AM3 and KmTx8, featuring both a hydrophilic polyol and a hydrophobic polyene chain, exhibited remarkable antifungal and antitumor activities.^{8,61,67} These pore-forming agents, taking a hairpin conformation in the lipid bilayer membrane, can specifically bind to membrane sterols and then perforate the membranes without altering the membrane integrity.^{63,67} It was demonstrated that AM3 directly interacts with the membrane cholesterol or ergosterol through the strict molecular recognition of the 3 β -OH group,⁸⁵ whereas KmTx8 was proposed to exhibit a cytotoxic mode-of-action against cancer cells through cholesterol and other membrane binding interactions.⁶⁷ Structure–activity relationship investigations of AM3 analogs revealed that the proper length of the polyol chains in these PPCs plays an important role in their membrane-disrupting activities. When the length of the polyol chains exceeds 35 carbons, the membrane-disrupting activity of the AM3 analog decreases significantly.⁵⁹ LFPs, such as ciguatoxins, brevetoxins, and maitotoxin, are activators of Na⁺ or Ca²⁺ channels in mammalian cell membranes.^{8,74–79} Although the target proteins and mode-of-action of maitotoxin remain unknown, its potent Ca²⁺ influx activity could be attributed to both hydrophilic and hydrophobic regions of this molecule. The *trans*-fused polycyclic ether feature of LFPs might be responsible for the toxicity of these marine toxins.⁸

The structure of benthol A (**1**), containing eight scattered ether rings on a C₇₂ backbone chain, is totally different from those of PPCs and LFPs. It cannot be divided into two separated hydrophilic and hydrophobic regions. In fact, our preliminary bioassays revealed that benthol A did not show either obvious antifungal activity against the model fungal species *Aspergillus niger* or significant cytotoxicity against eight human cancer cell

lines, *viz.*, A375, AGS, A549, A2780, A2780/T, HCT-8, HCT-8/T, and MDA-MB-231, at the concentration of 100.0 μ M. Evidently, the bioactivity profile of benthol A might be completely different from those of PPCs and LFPs. Most notably, in-depth mechanistic investigations of the potent antiparasitic activity of benthol A are warranted.

Marine dinoflagellates are exclusively focused on the generation of polyketides. From the standpoint of biosynthesis, SCCCs are unique and huge polyketides.⁸⁶ It is well known that small molecule polyketides, accounting for the majority of all the metabolites in bacteria, fungi, phytoplankton, plants, and invertebrates, are believed to be mainly synthesized by polyketide synthases. With the aid of modern techniques, the molecular bases of various polyketide synthases in bacteria and fungi have been thoroughly elucidated.^{87,88} However, large and complex genomes of dinoflagellates, and their limited genetic data have been hampering the molecular studies of SCCC biosynthesis. To date, the biosynthetic knowledge of SCCC has been accessed only by means of biochemical labeling studies.^{80,89} The biosynthetic investigation of SCCC through genome mining is still in the early stages.

Conclusions

In summary, an unprecedented SCCC, named benthol A, was isolated from the unexplored benthic dinoflagellate collected in the South China Sea. Its planar structure, containing a 6,6-spiroketol motif, three tetrahydrofuran rings, three tetrahydropyran moieties, and twenty-two hydroxy groups, was elucidated by means of extensive NMR spectroscopic investigations, particularly the 2D-INADEQUATE spectrum, acquired with ¹³C-enriched chemical entity. The absolute configurations of thirty-five carbon stereocenters were unambiguously established by a combination of periodate degradation of the 1,2-diol groups, ozonolysis of the carbon–carbon double bonds, JBCA, NOE interactions, modified Mosher's MTPA ester method, and DFT-NMR ¹³C chemical-shift calculations aided by DP4+ statistical analysis. Additionally, benthol A displayed potent antimalarial activity against *Plasmodium falciparum* 3D7 parasites. The fascinating structural features of benthol A place it as the first member of a new SCCC family located between PPCs and LFPs, herein termed polyol–polyether compounds. This suggestion was strongly supported by the PCA score plot of 188 SCCC reported to date. To our knowledge, this is the first report of a SCCC containing eight scattered ether rings. The discovery of benthol A does not only provide new insights into underexplored chemical space of SCCC, but also opens up a new window for skeletal diversity of SCCC. This work demonstrates that new taxa of dinoflagellates harbor unique biosynthetic machineries for the production of novel SCCC families with signature features.

Data availability

All experimental procedures, spectroscopic and computational data are provided as the ESI.



Author contributions

Conceptualization, J. W. and L. S.; methodology, W. H. C., T. K., M. H. Y., J. W., and L. S.; investigation, Z. P. J., S. H. S., Y. Y., A. M., and J. Y. L.; writing – original draft, S. H. S., A. M., J. Y. L., W. H. C., T. K., J. W., and L. S.; writing – review & editing, W. H. C., J. W., and L. S.; funding acquisition, T. K., J. W., and L. S.; resources, M. H. Y., T. K., J. W., and L. S.; supervision, T. K., J. W., and L. S.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 D. Uemura, *Chem. Rec.*, 2006, **6**, 235.
- 2 M. Kita and D. Uemura, *Chem. Rec.*, 2010, **10**, 48.
- 3 M. Murata, H. Naoki, T. Iwashita, S. Matsunaga, M. Sasaki, A. Yokoyama and T. Yasumoto, *J. Am. Chem. Soc.*, 1993, **115**, 2060.
- 4 T. Yasumoto and M. Murata, *Chem. Rev.*, 1993, **93**, 1897.
- 5 J. Kobayashi and M. Ishibashi, *Chem. Rev.*, 1993, **93**, 1753.
- 6 J. Kobayashi and T. Kubota, *J. Nat. Prod.*, 2007, **70**, 451.
- 7 S. A. Rasmussen, A. J. C. Andersen, N. G. Andersen, K. F. Nielsen, P. J. Hansen and T. O. Larsen, *J. Nat. Prod.*, 2016, **79**, 662.
- 8 T. Oishi, *Bull. Chem. Soc. Jpn.*, 2020, **93**, 1350.
- 9 M. Kita, N. Ohishi, K. Konishi, M. Kondo, T. Koyama, M. Kitamura, K. Yamada and D. Uemura, *Tetrahedron*, 2007, **63**, 6241.
- 10 H. Takamura, T. Fujiwara, Y. Kawakubo, I. Kadota and D. Uemura, *Chem.–Eur. J.*, 2016, **22**, 1979.
- 11 H. Takamura, T. Fujiwara, Y. Kawakubo, I. Kadota and D. Uemura, *Chem.–Eur. J.*, 2016, **22**, 1984.
- 12 H. Nakamura, T. Asari and A. Murai, *J. Am. Chem. Soc.*, 1995, **117**, 550.
- 13 W. S. Li, R. J. Yan, Y. Yu, Z. Shi, A. Mándi, L. Shen, T. Kurtán and J. Wu, *Angew. Chem. Int. Ed.*, 2020, **59**, 13028; *Angew. Chem.*, 2020, **132**, 13128.
- 14 W. S. Li, Z. Luo, Y. L. Zhu, Y. Yu, J. Wu and L. Shen, *Mar. Drugs*, 2020, **18**, 590.
- 15 M. S. Lee, D. J. Repeta, K. Nakanishi and M. G. Zagorski, *J. Am. Chem. Soc.*, 1986, **108**, 7855.
- 16 F. M. D. Ismail, L. Nahar and S. D. Sarker, *Phytochem. Anal.*, 2021, **32**, 7.
- 17 Y. Hamamoto, K. Tachibana, P. T. Holland, F. Shi, V. Beuzenberg, Y. Itoh and M. Satake, *J. Am. Chem. Soc.*, 2012, **134**, 4963.
- 18 N. Matsumori, D. Kaneno, M. Murata, H. Nakamura and K. Tachibana, *J. Org. Chem.*, 1999, **64**, 866.
- 19 G. Bifulco, P. Dambruoso, L. Gomez-Paloma and R. Riccio, *Chem. Rev.*, 2007, **107**, 3744.
- 20 I. Ohtani, T. Kusumi, Y. Kashman and H. Kakisawa, *J. Am. Chem. Soc.*, 1991, **113**, 4092.
- 21 F. Freire, J. M. Seco, E. Quiñoá and R. Riguera, *J. Org. Chem.*, 2005, **70**, 3778.
- 22 J. M. Seco, E. Quiñoá and R. Riguera, *Chem. Rev.*, 2012, **112**, 4603.
- 23 K. Konno, T. Fujishima and Z. P. Liu, *Chirality*, 2002, **13**, 72.
- 24 H. Takamura, Y. Kadonaga, Y. Yamano, C. Han, Y. Aoyama, I. Kadota and D. Uemura, *Tetrahedron Lett.*, 2009, **50**, 863.
- 25 C. Han, Y. Yamano, M. Kita, H. Takamura and D. Uemura, *Tetrahedron Lett.*, 2009, **50**, 5280.
- 26 H. Takamura, Y. Kadonaga, Y. Yamano, C. Han, I. Kadota and D. Uemura, *Tetrahedron*, 2009, **65**, 7449.
- 27 H. Takamura, Y. Kadonaga, I. Kadota and D. Uemura, *Tetrahedron*, 2010, **66**, 7569.
- 28 E. Van der Eycken, H. D. Wilde, L. Deprez and M. Vandewalle, *Tetrahedron Lett.*, 1987, **28**, 4759.
- 29 V. K. Tandon, A. M. Van Leusen and H. Wynberg, *J. Org. Chem.*, 1983, **48**, 2767.
- 30 S. G. Smith and J. M. Goodman, *J. Am. Chem. Soc.*, 2010, **132**, 12946.
- 31 N. Grimblat, M. M. Zanardi and A. M. Sarotti, *J. Org. Chem.*, 2015, **80**, 12526.
- 32 W. S. Li, A. Mándi, J. J. Liu, L. Shen, T. Kurtán and J. Wu, *J. Org. Chem.*, 2019, **84**, 2596.
- 33 A. Mándi and T. Kurtán, *Nat. Prod. Rep.*, 2019, **36**, 889.
- 34 C. Adamo and V. Barone, *J. Chem. Phys.*, 1998, **108**, 664.
- 35 S. Qiu, E. De Gussem, K. A. Tehrani, S. Sergeyev, P. Bultinck and W. Herrebout, *J. Med. Chem.*, 2013, **56**, 8903.
- 36 W. S. Wang, J. M. Yao, Z. Chen, Y. M. Sun, Y. Q. Shi, Y. F. Wei, H. J. Zhou, Y. F. Yu, S. Z. Li and L. P. Duan, *Br. J. Pharmacol.*, 2020, **177**, 5569.
- 37 J. L. Zhao, J. M. Feng, Z. Tan, J. M. Liu, J. Y. Zhao, R. D. Chen, K. B. Xie, D. W. Zhang, Y. Li, L. Y. Yu, X. G. Chen and J. G. Dai, *J. Nat. Prod.*, 2017, **80**, 1819.
- 38 J. L. Ren, X. P. Zou, W. S. Li, L. Shen and J. Wu, *Mar. Drugs*, 2018, **16**, 434.
- 39 M. Satake, M. Murata, T. Yasumoto, T. Fujita and H. Naoki, *J. Am. Chem. Soc.*, 1991, **113**, 9859.
- 40 G. K. Paul, N. Matsumori, M. Murata and K. Tachibana, *Tetrahedron Lett.*, 1995, **36**, 6279.
- 41 G. K. Paul, N. Matsumori, K. Konoki, M. Murata and K. Tachibana, *J. Mar. Biotechnol.*, 1997, **5**, 124.



- 42 Y. Doi, M. Ishibashi, H. Nakamichi, T. Kosaka, T. Ishikawa and J. Kobayashi, *J. Org. Chem.*, 1997, **62**, 3820.
- 43 T. Kubota, M. Tsuda, Y. Doi, A. Takahashi, H. Nakamichi, M. Ishibashi, E. Fukushi, J. Kawabata and J. Kobayashi, *Tetrahedron*, 1998, **54**, 14455.
- 44 M. Murata, S. Matsuoka, N. Matsumori, G. K. Paul and K. Tachibana, *J. Am. Chem. Soc.*, 1999, **121**, 870.
- 45 T. Houdai, S. Matsuoka, M. Murata, M. Satake, S. Ota, Y. Oshima and L. L. Rhodes, *Tetrahedron*, 2001, **57**, 5551.
- 46 X. Huang, D. Zhao, Y. Guo, H. Wu, L. Lin, Z. Wang, J. Ding and Y. Lin, *Bioorg. Med. Chem. Lett.*, 2004, **14**, 3117.
- 47 X. Huang, D. Zhao, Y. Guo, H. Wu, E. Trivellone and G. Cimino, *Tetrahedron Lett.*, 2004, **45**, 5501.
- 48 N. Morsy, S. Matsuoka, T. Houdai, N. Matsumori, S. Adachi, M. Murata, T. Iwashita and T. Fujita, *Tetrahedron*, 2005, **61**, 8606.
- 49 T. Kubota, A. Takahashi, M. Tsuda and J. Kobayashi, *Mar. Drugs*, 2005, **3**, 113.
- 50 R. Echigoya, L. Rhodes, Y. Oshima and M. Satake, *Harmful Algae*, 2005, **4**, 383.
- 51 K. Washida, T. Koyama, K. Yamada, M. Kita and D. Uemura, *Tetrahedron Lett.*, 2006, **47**, 2521.
- 52 N. Morsy, T. Houdai, S. Matsuoka, N. Matsumori, S. Adachi, T. Oishi, M. Murata, T. Iwashita and T. Fujita, *Bioorg. Med. Chem.*, 2006, **14**, 6548.
- 53 S. Huang, C. Kuo, Y. Lin, Y. Chen and C. Lu, *Tetrahedron Lett.*, 2009, **50**, 2512.
- 54 Y. Meng, R. M. Van Wagoner, I. Misner, C. Tomas and J. L. C. Wright, *J. Nat. Prod.*, 2010, **73**, 409.
- 55 N. Hanif, O. Ohno, M. Kitamura, K. Yamada and D. Uemura, *J. Nat. Prod.*, 2010, **73**, 1318.
- 56 T. Inuzuka, Y. Yamamoto, K. Yamada and D. Uemura, *Tetrahedron Lett.*, 2012, **53**, 239.
- 57 G. Nuzzo, A. Cutignano, A. Sardo and A. Fontana, *J. Nat. Prod.*, 2014, **77**, 1524.
- 58 T. Inuzuka, K. Yamada and D. Uemura, *Tetrahedron Lett.*, 2014, **55**, 6319.
- 59 M. Satake, K. Cornelio, S. Hanashima, R. Malabed, M. Murata, N. Matsumori, H. Zhang, F. Hayashi, S. Mori, J. S. Kim, C. H. Kim and J. S. Lee, *J. Nat. Prod.*, 2017, **80**, 2883.
- 60 A. Cutignano, G. Nuzzo, A. Sardo and A. Fontana, *Mar. Drugs*, 2017, **15**, 157.
- 61 Y. Wakamiya, M. Ebine, M. Murayama, H. Omizu, N. Matsumori, M. Murata and T. Oishi, *Angew. Chem. Int. Ed.*, 2018, **57**, 6060; *Angew. Chem.*, 2018, **130**, 6168.
- 62 K. A. Martinez, C. Lauritano, D. Druka, G. Romano, T. Grohmann, M. Jaspars, J. Martin, C. Diaz, B. Cautain, M. de la Cruz, A. Ianora and F. Reyes, *Mar. Drugs*, 2019, **17**, 385.
- 63 Y. Wakamiya, M. Ebine, N. Matsumori and T. Oishi, *J. Am. Chem. Soc.*, 2020, **142**, 3472.
- 64 R. M. Van Wagoner, J. R. Deeds, M. Satake, A. A. Ribeiro, A. R. Place and J. L. C. Wright, *Tetrahedron Lett.*, 2008, **49**, 6457.
- 65 J. Peng, A. R. Place, W. Yoshida, C. Anklin and M. T. Hamann, *J. Am. Chem. Soc.*, 2010, **132**, 3277.
- 66 R. M. Van Wagoner, J. R. Deeds, A. O. Tatters, A. R. Place, C. R. Tomas and J. L. C. Wright, *J. Nat. Prod.*, 2010, **73**, 1360.
- 67 A. L. Waters, J. Oh, A. R. Place and M. T. Hamann, *Angew. Chem. Int. Ed.*, 2015, **54**, 15705; *Angew. Chem.*, 2015, **127**, 15931.
- 68 P. Cai, S. He, C. Zhou, A. R. Place, S. Had, L. Ding, H. Chen, Y. Jiang, C. Guo, Y. Xu, J. Zhang and X. Yan, *Harmful Algae*, 2016, **58**, 66.
- 69 S. A. Rasmussen, S. B. Binzer, C. Hoeck, S. Meier, L. S. de Medeiros, N. G. Andersen, A. Place, K. F. Nielsen, P. J. Hansen and T. O. Larsen, *J. Nat. Prod.*, 2017, **80**, 1287.
- 70 B. S. Hwang, E. Y. Yoon, H. S. Kim, W. Yih, J. Y. Park, H. J. Jeong and J. R. Rho, *Bioorg. Med. Chem. Lett.*, 2013, **23**, 3023.
- 71 B. S. Hwang, E. Y. Yoon, E. J. Jeong, J. Park, E. H. Kim and J. R. Rho, *J. Org. Chem.*, 2018, **83**, 194.
- 72 K. Sugahara, Y. Kitamura, M. Murata, M. Satake and K. Tachibana, *J. Org. Chem.*, 2011, **76**, 3131.
- 73 H. J. Domínguez, D. Cabrera-García, C. Cuadrado, A. Novelli, M. T. Fernández-Sánchez, J. J. Fernández and A. H. Daranas, *Org. Lett.*, 2021, **23**, 13.
- 74 M. Murata, A. M. Legrand, Y. Ishibashi, M. Fukui and T. Yasumoto, *J. Am. Chem. Soc.*, 1990, **112**, 4380.
- 75 K. Yogi, N. Oshiro, Y. Inafuku, M. Hiramata and T. Yasumoto, *Anal. Chem.*, 2011, **83**, 8886.
- 76 J. R. Chekan, T. R. Fallon and B. S. Moore, *Curr. Opin. Chem. Biol.*, 2020, **59**, 119.
- 77 Y. Y. Lin, M. Risk, S. M. Ray, D. Van Engen, J. Clardy, J. Golik, J. C. James and K. Nakanishi, *J. Am. Chem. Soc.*, 1981, **103**, 6773.
- 78 K. Tanaka, Y. Itagaki, M. Satake, H. Naoki, T. Yasumoto, K. Nakanishi and N. Berova, *J. Am. Chem. Soc.*, 2005, **127**, 9561.
- 79 R. Suzuki, R. Irie, Y. Harntaweessup, K. Tachibana, P. T. Holland, D. T. Harwood, F. Shi, V. Beuzenberg, Y. Itoh, S. Pascal, P. J. B. Edwards and M. Satake, *Org. Lett.*, 2014, **16**, 5850.
- 80 M. Yamazaki, M. Izumikawa, K. Tachibana, M. Satake, Y. Itoh and M. Hashimoto, *J. Org. Chem.*, 2012, **77**, 4902.
- 81 P. Ciminiello, E. Fattorusso, M. Forino, R. Poletti and R. Viviani, *Chem. Res. Toxicol.*, 2000, **13**, 770.
- 82 G. Ivosev, L. Burton and R. Bonner, *Anal. Chem.*, 2008, **80**, 4933.
- 83 M. Yu, C. Zhou, D. Tian, H. M. Jia, Z. Q. Li, C. Yang, Y. M. Ba, H. K. Wu and Z. M. Zou, *J. Pharm. Biomed. Anal.*, 2021, **198**, 114004.
- 84 K. Onodera, H. Nakamura, Y. Oba, Y. Ohizumi and M. Ojika, *J. Am. Chem. Soc.*, 2005, **127**, 10406.
- 85 R. A. Espiritu, N. Matsumori, M. Tsuda and M. Murata, *Biochemistry*, 2014, **53**, 3287.
- 86 R. M. Van Wagoner, M. Satake and J. L. C. Wright, *Nat. Prod. Rep.*, 2014, **31**, 1101.
- 87 J. Clardy and C. Walsh, *Nature*, 2004, **432**, 829.
- 88 A. Nivina, K. P. Yuet, J. Hsu and C. Khosla, *Chem. Rev.*, 2019, **119**, 12524.
- 89 S. L. MacKinnon, A. D. Cembella, I. W. Burton, N. Lewis, P. LeBlanc and J. A. Walter, *J. Org. Chem.*, 2006, **71**, 8724.

