


AUTHOR QUERY FORM

	<p>Journal: J. Appl. Phys.</p> <p>Article Number: 068429JAP</p>	<p>Please provide your responses and any corrections by annotating this PDF and uploading it according to the instructions provided in the proof notification email.</p>
---	--	--

Dear Author,

Below are the queries associated with your article; please answer all of these queries before sending the proof back to AIP. Please indicate the following:

Figures that are to appear as color online only (i.e., Figs. 1, 2, 3) _____ (this is a free service).

Figures that are to appear as color online and color in print _____ (a fee of \$325 per figure will apply).

Article checklist: In order to ensure greater accuracy, please check the following and make all necessary corrections before returning your proof.

1. Is the title of your article accurate and spelled correctly?
2. Are the author names in the proper order and spelled correctly?
3. Please check affiliations including spelling, completeness, and correct linking to authors.
4. Did you remember to include acknowledgment of funding, if required, and is it accurate?

Location in article	Query / Remark: click on the Q link to navigate to the appropriate spot in the proof. There, insert your comments as a PDF annotation.
AQ1	Please check the edit made in affiliations 1 and 3.
AQ2	Please check the section hierarchy.
AQ3	Refs. 23 and 25 contain identical information. Please check and provide the correct reference or delete the duplicate reference. If the duplicate is deleted, renumber the reference list as needed and update all citations in the text.

Thank you for your assistance.

1 Influence of the substrate choice on the $L1_0$ phase formation 2 of post-annealed Pt/Fe and Pt/Ag/Fe thin films

3 I. A. Vladymyrskiy,^{1,a)} M. V. Karpets,² G. L. Katona,³ D. L. Beke,³ S. I. Sidorenko,¹
4 T. Nagata,⁴ T. Nabatame,⁴ T. Chikyow,⁴ F. Ganss,⁵ G. Beddies,⁵ M. Albrecht,^{5,6}
5 and I. M. Makogon¹

6 ¹*Metal Physics Department, National Technical University of Ukraine "KPI," Prospect Peremogy 37,*
7 *03056 Kyiv, Ukraine*

8 ²*Frantsevich Institute for Problems of Materials Science, National Academy of Sciences of Ukraine,*
9 *3 Krzhizhanovsky Str., 03142 Kyiv, Ukraine*

10 ³*Department of Solid State Physics, University of Debrecen, P.O. Box 2, H-4010 Debrecen, Hungary*

11 ⁴*MANA Foundry and MANA Advanced Device Materials Group, National Institute for Materials Science,*
12 *1-1 Namiki, Tsukuba 305-0044, Japan*

13 ⁵*Institute of Physics, Chemnitz University of Technology, D-09107 Chemnitz, Germany*

14 ⁶*Institute of Physics, University of Augsburg, Universitätsstraße 1, D-86159 Augsburg, Germany*

15 (Received 29 May 2014; accepted 16 July 2014; published online xx xx xxxx)

16 Pt/Fe and Pt/Ag/Fe layered films were deposited by DC magnetron sputtering on MgO(001),
17 SrTiO₃(001), and Al₂O₃(0001) single crystalline substrates at room temperature. The films were
18 post-annealed between 623 K and 1173 K for 30 s in flowing N₂ atmosphere. The onset of the
19 $L1_0$ -FePt phase formation in films deposited on MgO(001) and SrTiO₃(001) substrates was
20 observed after annealing between 773 and 873 K, while chemical $L1_0$ ordering sets in for Pt/Fe
21 bilayers on Al₂O₃(0001) at lower temperatures accompanied by strong (001)-texture. It is
22 concluded that elastic stress, arising from the difference in thermal expansion coefficients between
23 film and substrate, promotes ordering and texture formation. © 2014 AIP Publishing LLC.

[<http://dx.doi.org/10.1063/1.4891477>]

AQ2 24 I. INTRODUCTION

25 FePt thin films have attracted considerable attention as
26 potential ultra-high density magnetic recording material
27 because of the high magneto-crystalline anisotropy of the
28 chemically ordered $L1_0$ -FePt phase.¹⁻⁵ These films also show
29 high saturation magnetization and excellent corrosion resist-
30 ance.⁶ Typically, the $L1_0$ -FePt phase forms from the disor-
31 dered A1-FePt phase after post-annealing or after deposition
32 onto heated substrates. However, industrial application of
33 FePt thin films requires chemical ordering at low temperatures
34 and the control of grain size and orientation.^{2,6-8} The ordering
35 temperature can be reduced by Fe/Pt multilayer deposition
36 followed by post-annealing^{9,10} or by introducing third ele-
37 ments such as Ag, Au, and Cu.¹¹⁻¹⁶ A pronounced (001)-tex-
38 ture in $L1_0$ -FePt films can be achieved by epitaxial growth at
39 elevated temperatures on single crystalline substrates such as
40 MgO(001)¹⁷⁻¹⁹ or SrTiO₃(001).²⁰ Furthermore, elastic stress
41 during rapid thermal annealing of FePt films on suitable sub-
42 strates is the origin of strain favoring the growth of (001)-ori-
43 ented grains in $L1_0$ ordered films.^{8,21,22}

44 In this study, the influence of various single crystalline
45 substrates (MgO(001), SrTiO₃(001), and Al₂O₃(0001)) on the
46 structural properties of post-annealed Pt/Fe and Pt/Ag/Fe thin
47 films was investigated using various techniques, including
48 secondary neutral mass spectrometry (SNMS). It is expected
49 that differences in the structural properties of the films after
50 post-annealing arise from the stress created by the difference

in thermal expansion coefficients of the metallic film and the
substrates (Al₂O₃(0001): $10 \times 10^{-6} \text{ K}^{-1}$, SrTiO₃(001): $9.4 \times$
 10^{-6} K^{-1} , and MgO(001): $5 \times 10^{-6} \text{ K}^{-1}$ (Ref. 8)).

54 II. EXPERIMENTAL

55 Layered films of Pt(15 nm)/Ag(0; 10 nm)/Fe(15 nm) were
56 deposited at room temperature on MgO(001), SrTiO₃(001), and
57 Al₂O₃(0001) single crystalline substrates by DC magnetron sput-
58 tering using individual Pt, Ag, and Fe targets. The Ar pressure in
59 the sputtering chamber was adjusted to 0.48 Pa for all deposi-
60 tions. The nominal thicknesses of the layers were evaluated
61 from the sputtering time (determined from the calibrated deposi-
62 tion rate of each target) and verified by profiler measurements.
63 Post-annealing of the films up to 1173 K was carried out in flow-
64 ing N₂ atmosphere (with 0.2 l/min flowing speed) for 30 s, using
65 a constant heating rate of 10 K/s. The structure of the films was
66 analyzed with an x-ray diffractometer (XRD) equipped with 2-
67 dimensional (2D) and scintillation detectors using Cu K_α radi-
68 ation. Composition-time (depth) profiles of post-annealed sam-
69 ples were determined by SNMS, using a low-pressure radio-
70 frequency Ar plasma both as source for ion bombardment and as
71 post-ionization medium.^{23,24} Furthermore, the magnetic proper-
72 ties were measured by superconductive quantum interference
73 device-vibrating sample magnetometry (SQUID-VSM).

74 III. RESULTS

75 A. Pt/Fe and Pt/Ag/Fe films on MgO(001)

76 Fig. 1 shows 2D XRD images of Pt/Fe and Pt/Ag/Fe films
77 sputter-deposited on MgO(001) substrates after post-annealing

^{a)}Author to whom correspondence should be addressed. Electronic mail:
vladymyrskiy@kpm.kpi.ua.

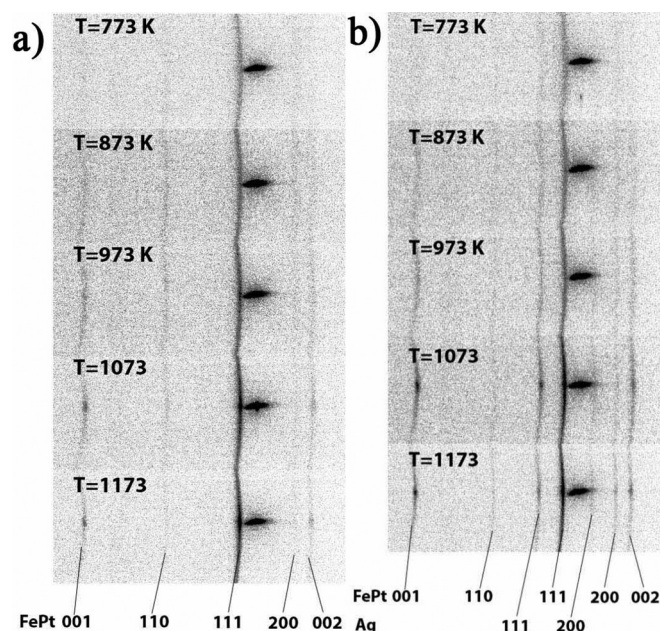


FIG. 1. 2D-XRD images of (a) Pt/Fe and (b) Pt/Ag/Fe films on MgO(001) substrates after post-annealing at different temperatures.

texture.⁶ However, in our case, the ordering mechanism is quite different as bilayer Fe/Pt films were deposited at room temperature and then post-annealed. Thus, FePt ordering is initiated in the Fe/Pt bilayer apart from the substrate.

SNMS composition profiles versus sputtering time of Pt/Fe and Pt/Ag/Fe films after post-annealing at 773 K and 873 K are presented in Fig. 2. Please note that the composition profile was calculated assuming a linear dependence of the measured intensities on the elemental concentration.²⁵ Post-annealing of the Pt/Fe film at 773 K leads to an almost homogeneous intermixing between the Pt and Fe layers. Further increase of the temperature up to 873 K does not change significantly the concentration profile. Post-annealing of the Pt/Ag/Fe film at 773 K also leads to the formation of a homogeneous FePt layer and to a moderate penetration of Ag into the FePt layer with an Ag rich layer on the top surface. Please note that from the individual Fe and Pt layer thicknesses, a composition of Fe₅₇Pt₄₃ is expected.

B. Pt/Fe and Pt/Ag/Fe films on SrTiO₃(001)

Fig. 3 shows 2D XRD images of Pt/Fe and Pt/Ag/Fe films sputter-deposited on SrTiO₃(001) substrates after post-annealing at temperatures up to 1173 K. Please note that the (100) reflection of the SrTiO₃ substrate and the (001) reflection of L₁₀-FePt are superimposed and cannot be easily distinguished. However, the onset of L₁₀ chemical ordering was registered after post-annealing between 773 K and 873 K in both Pt/Fe and Pt/Ag/Fe films. Intensity of superstructure reflection becomes stronger with higher annealing temperatures. It is apparent that after post-annealing at 1073 K and 1173 K, a non-uniform distribution of the (001) peak intensity along the diffraction ring is observed, revealing that some part of the grains are preferentially oriented along the [001] direction. But the strong (111) peak is still present in the XRD images.

The SNMS depth profiles of post-annealed films grown on SrTiO₃(001) substrates (Fig. 4) are very similar to the

at various temperatures. The appearance of the (001) superstructure peak for both Pt/Fe and Pt/Ag/Fe films was weakly indicated after annealing at 773 K but clearly observed after annealing at 873 K, indicating the onset of the L₁₀-FePt phase formation. With increasing annealing temperature, this reflection increases in intensity and becomes sharper. However, the intensity of the (111) peak is still much higher than the intensity of (001) reflection, indicating the presence of strong (111)-texture. Furthermore, the intensity of the (111) Ag reflection observed for the Pt/Ag/Fe films also becomes more pronounced with increasing temperature. Please note that it is well known that FePt films grown on MgO(001) substrates at elevated temperatures reveal high L₁₀ ordering with pronounced (001)

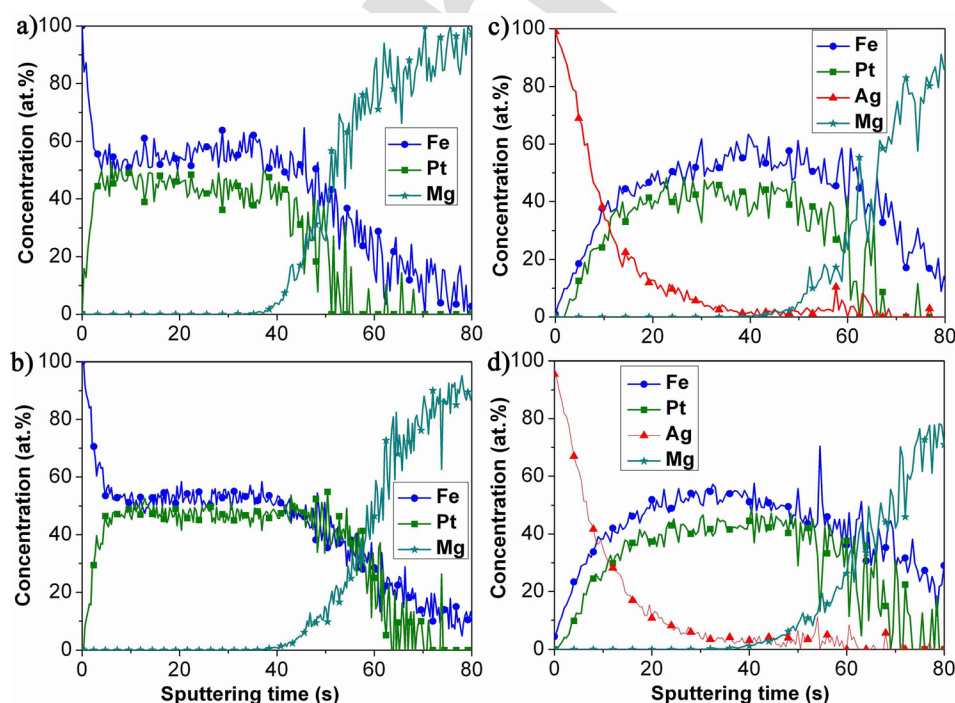


FIG. 2. Composition profiles of films on MgO(001) substrates after post-annealing at 773 K and 873 K: (a) Pt/Fe (773 K); (b) Pt/Fe (873 K); (c) Pt/Ag/Fe (773 K); and (d) Pt/Ag/Fe (873 K).

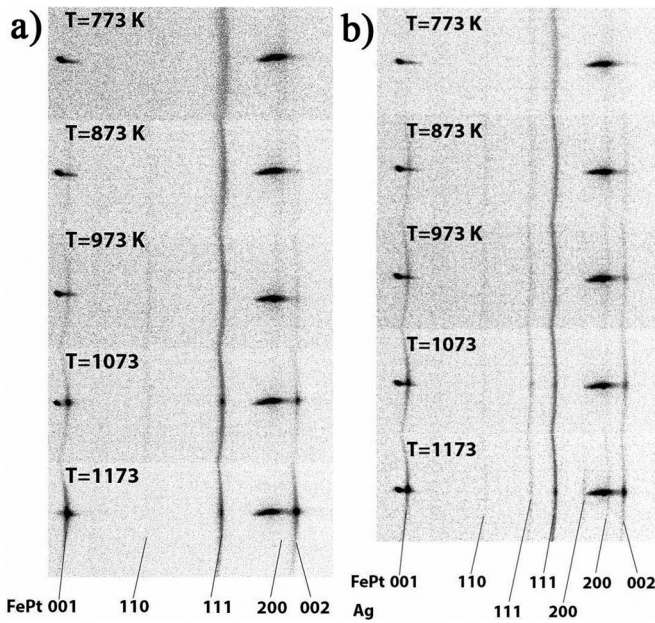


FIG. 3. 2D-XRD images of (a) Pt/Fe and (b) Pt/Ag/Fe films on SrTiO₃(001) substrates after post-annealing at different temperatures.

(001) superstructure peak is present on the XRD images even after annealing at 773 K for both the Pt/Fe and Pt/Ag/Fe films. These peaks have low intensity as compared to the fundamental (111) reflection, but the intensity is not uniformly distributed along the diffraction ring and has a well defined maximum on the equatorial line, indicating the onset of (001)-texture formation. Additional XRD measurements showed the appearance of the low intensity (001) reflection for the Pt/Fe film after post-annealing at 623 K and for the Pt/Ag/Fe film after post-annealing at 673 K (Fig. 6). The intensity of the (001) reflection increases drastically with increasing annealing temperature, confirming the pronounced (001)-texture for samples post-annealing at 1073 K, which is slightly less pronounced in Pt/Ag/Fe films. However, in Pt/Ag/Fe, a strong (111) reflection remains.

Fig. 7 shows the corresponding SNMS concentration depth profiles of the Pt/Fe and Pt/Ag/Fe films after post-annealing at 773 K and 873 K. The results are very similar to those obtained for films deposited on MgO(001) and SrTiO₃(001) substrates. Even after annealing at 773 K, there is an almost homogeneous distribution of Pt and Fe in the Fe/Pt film. Increase of the annealing temperature does not lead to a substantial modification of the concentration profile. For the Pt/Ag/Fe film, full intermixing of the Fe and Pt layers with pronounced Ag surface segregation after post-annealing at 773 K was obtained.

The structural analysis of FePt films formed on Al₂O₃(0001) after post-annealing revealed L1₀ ordering and a pronounced (001)-texture, thus a strong perpendicular magnetic anisotropy might be expected in these films. The magnetic properties of the annealed films were investigated by SQUID-VSM. M-H hysteresis loops were measured at room temperature in two geometries: magnetic field is applied in the film plane and out of the film plane. Figure 8 shows normalized M-H hysteresis loops obtained for Pt/Fe and Pt/Ag/Fe films after post-annealing at high temperatures

profiles obtained on MgO(001) substrates. Also in this case, annealing at 773 K leads to almost full intermixing between the Fe and Pt layers. Increase of the annealing temperature does not lead to significant changes in the composition distribution. Annealing of films with Ag intermediate layer causes again a moderate penetration of Ag into the FePt layer and segregation towards the free surface.

C. Pt/Fe and Pt/Ag/Fe films on Al₂O₃(0001)

Fig. 5 shows the 2D XRD images of Pt/Fe and Pt/Ag/Fe films sputter-deposited on Al₂O₃(0001) substrates after post-annealing at various temperatures. It is apparent that the

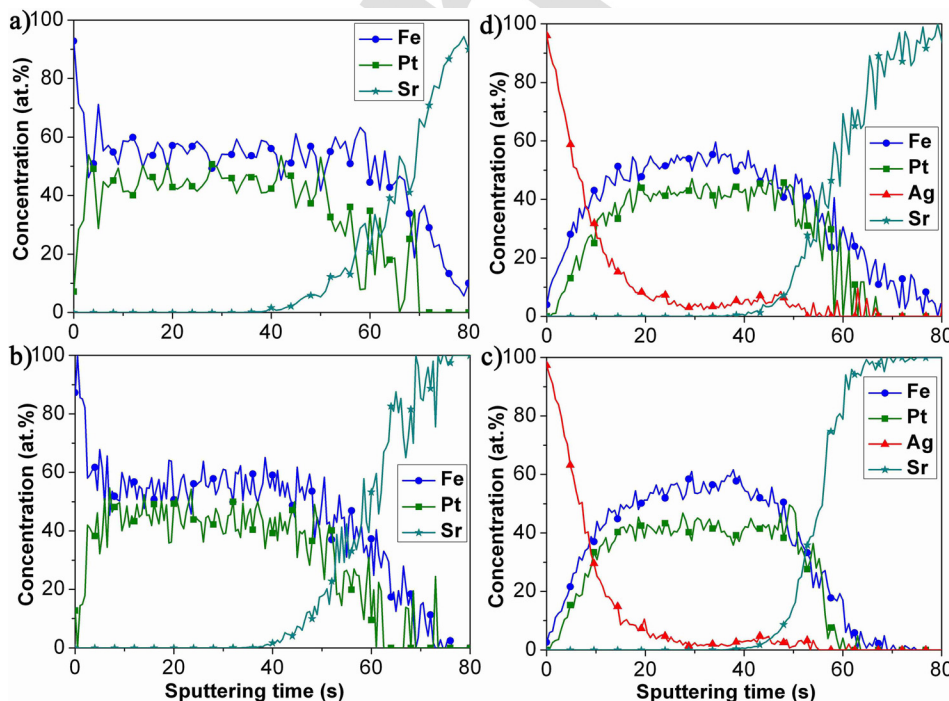


FIG. 4. Composition profiles of films on SrTiO₃(001) substrates after post-annealing at 773 K and 873 K: (a) Pt/Fe (773 K); (b) Pt/Fe (873 K); (c) Pt/Ag/Fe (773 K); and (d) Pt/Ag/Fe (873 K).

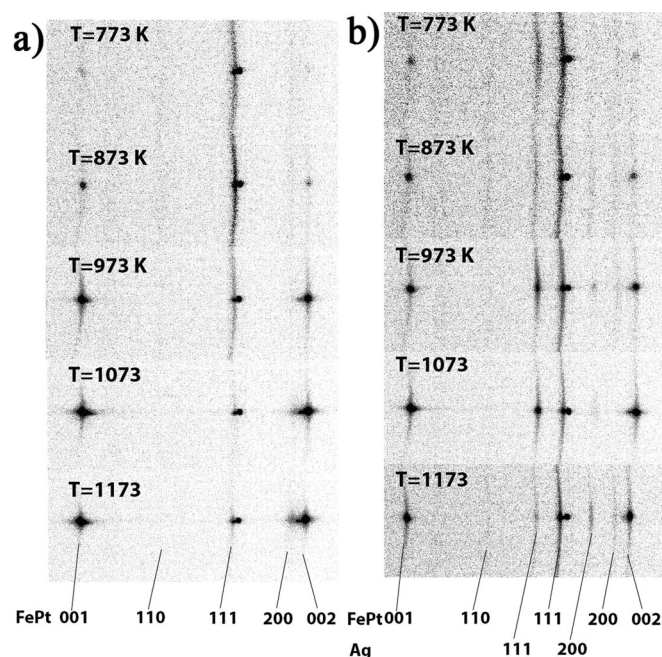


FIG. 5. 2D-XRD images of (a) Pt/Fe and (b) Pt/Ag/Fe films on $\text{Al}_2\text{O}_3(0001)$ substrates after post-annealing at different temperatures.

173 (973 K–1073 K). The magnetization curves for all films are
 174 quite similar and a more or less isotropic behavior in mag-
 175 netization reversal for the in-plane and out-of-plane field
 176 directions is observed. The coercivity of the Pt/Fe films after
 177 post-annealing at 973 K and 1073 K is 14.0 kOe and 15.8
 178 kOe, respectively. For films with Ag intermediate layer,
 179 these values were increased up to 17.7 kOe and 24.2 kOe,
 180 respectively. This behavior can be explained by Ag diffusion
 181 to the grain boundaries, which results in exchange decou-
 182 pling of FePt grains, which in turn enhances the coercivity.

183 Furthermore, the still present (111) and (200) orientations
 184 are responsible for the isotropic magnetic properties due to
 185 the rather randomly oriented FePt grains.

186 XRD results presented above indicate that the deposition
 187 of Pt/Fe and Pt/Ag/Fe layered films onto single crystalline
 188 $\text{Al}_2\text{O}_3(0001)$ substrates with hexagonal structure leads to
 189 decrease of the $L1_0$ -FePt phase formation temperature as
 190 compared to $\text{MgO}(001)$ and $\text{SrTiO}_3(001)$ substrates with
 191 cubic lattice. Moreover, annealing of the films on
 192 $\text{Al}_2\text{O}_3(0001)$ substrate results in pronounced (001)-
 193 texture formation. On the other hand, introduction of the Ag
 194 intermediate layer leads to the slight deterioration of the (001)-
 195 texture with the presence of a strong (111) reflection. Results
 196 of the SNMS depth profiling were very similar for films sput-
 197 tered onto the all investigated single crystalline substrates:
 198 even after post-annealing at 773 K almost homogeneous
 199 intermixing of the Pt and Fe layers was observed. This fact
 200 indicates that there is no noticeable effect of the substrate
 201 type on the diffusion processes but its influence on the chem-
 202 ical ordering and texture formation is more significant.

203 The Ag intermediate layer increases the coercivity of
 204 the films after their post-annealing. This can be explained by
 205 decreasing the magnetic interaction between the $L1_0$ -FePt
 206 grains due to their isolation. Isolated grains were formed
 207 because of the limited Ag solubility in FePt lattice and its
 208 grain boundary and surface segregation tendency (as was
 209 shown above). This conclusion is in agreement with the con-
 210 clusions obtained in Ref. 26. Despite the pronounced (001)-
 211 texture, films deposited onto $\text{Al}_2\text{O}_3(0001)$ substrates are
 212 magnetically isotropic, indicating the presence of chemically
 213 disordered A1 grains and $L1_0$ ordered grains with (111) and
 214 (200) orientations.

215 The observed differences in the texture formation in the
 216 films deposited onto different substrates can be explained by

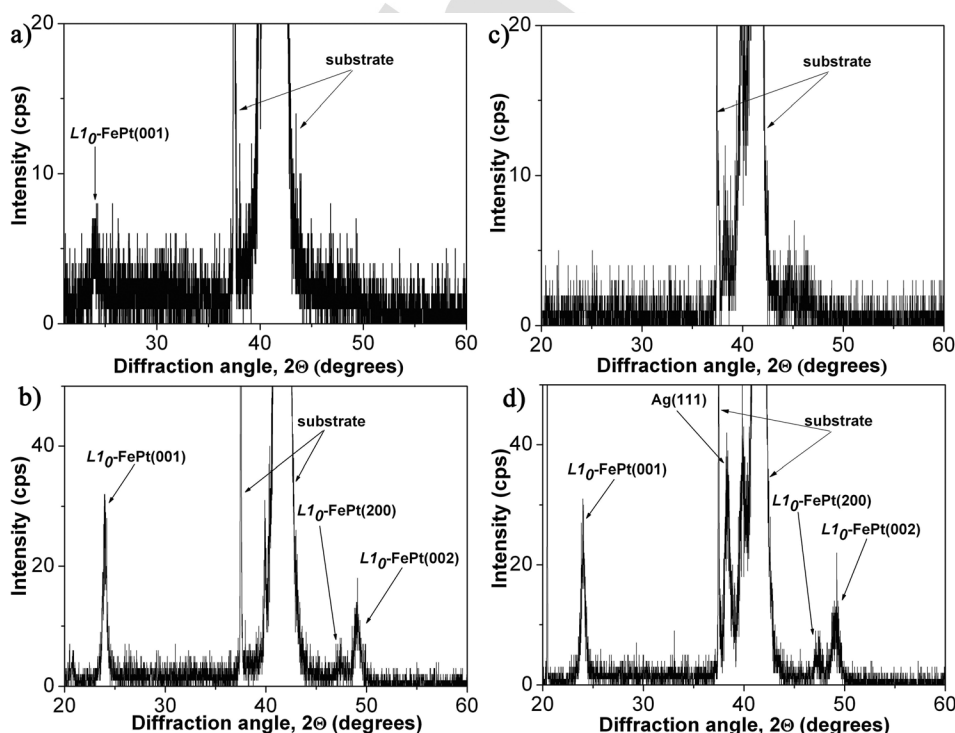


FIG. 6. XRD (θ - 2θ)-scans of film samples deposited on $\text{Al}_2\text{O}_3(0001)$ substrates after post annealing at different temperatures: (a) Pt/Fe at 623 K; (b) Pt/Fe at 673 K; (c) Pt/Ag/Fe at 623 K; and (d) Pt/Ag/Fe at 673 K.

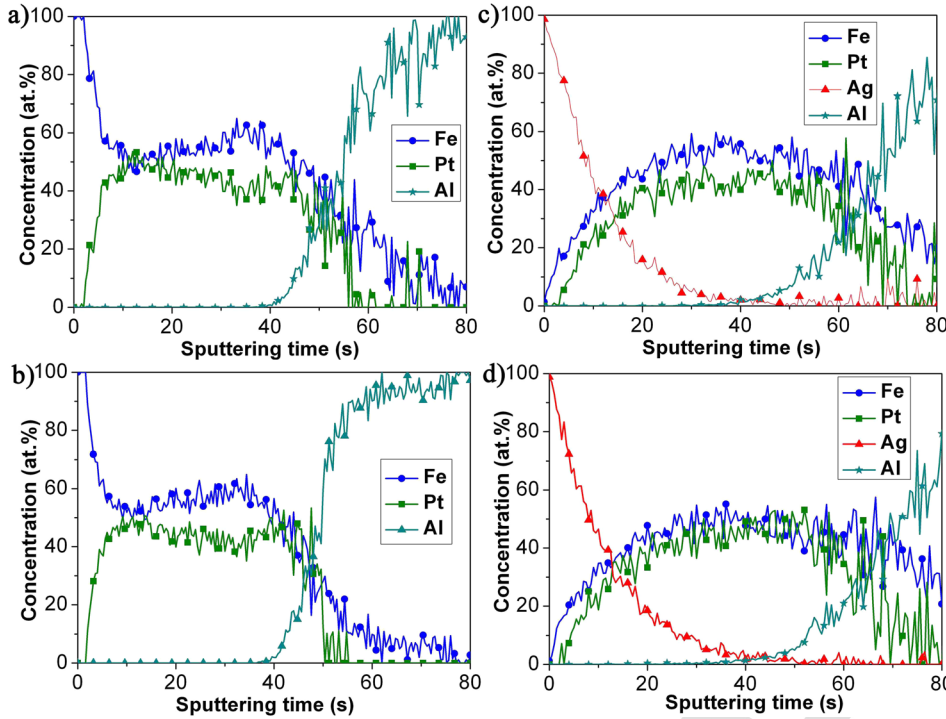


FIG. 7. Composition profiles of films on $\text{Al}_2\text{O}_3(0001)$ substrates after post-annealing at different temperatures: (a) Pt/Fe at 773 K; (b) Pt/Fe at 873 K; (c) Pt/Ag/Fe at 773 K; and (d) Pt/Ag/Fe at 873 K.

217 stresses, created by the mismatch between crystal lattices
 218 and by the difference of the thermal expansion coefficients
 219 of metallic layers and the substrate. These stresses are the origin
 220 of the strain favoring the growth of [001]-oriented
 221 grains.²² Stresses arising due to the difference in thermal
 222 expansion coefficients can be calculated from the equation²⁷

$$\sigma = \Delta\alpha\Delta TE / (1 - \mu),$$

223 where $\Delta\alpha$ is the difference in the thermal expansion coefficients
 224 between the substrate and the film, ΔT is the temperature
 225 difference between room and the post-annealing
 226 temperatures, E is the elastic modulus of the film, and μ is

Poisson's ratio. As the elastic modulus is temperature dependent,
 227 and in our case the Fe layer interacts with the Pt,
 228 forming first $A1\text{-FePt}$ and then $L1_0\text{-FePt}$ phases, we used the
 229 elastic modulus (180 GPa), Poisson's ratio (0.33), and thermal
 230 expansion coefficient ($10.5 \times 10^{-6} \text{ K}^{-1}$) for bulk FePt
 231 for the estimation of the difference in the stresses occurring
 232 in the films deposited onto different substrates. Fig. 9 shows
 233 the calculated stresses that arise from thermal expansion mismatch
 234 as a function of temperature for the investigated samples. It is clear
 235 that the level of compressive stresses in the films deposited onto
 236 $\text{Al}_2\text{O}_3(0001)$ substrates is much higher as compared to
 237 $\text{MgO}(001)$ and $\text{SrTiO}_3(001)$ substrates.
 238

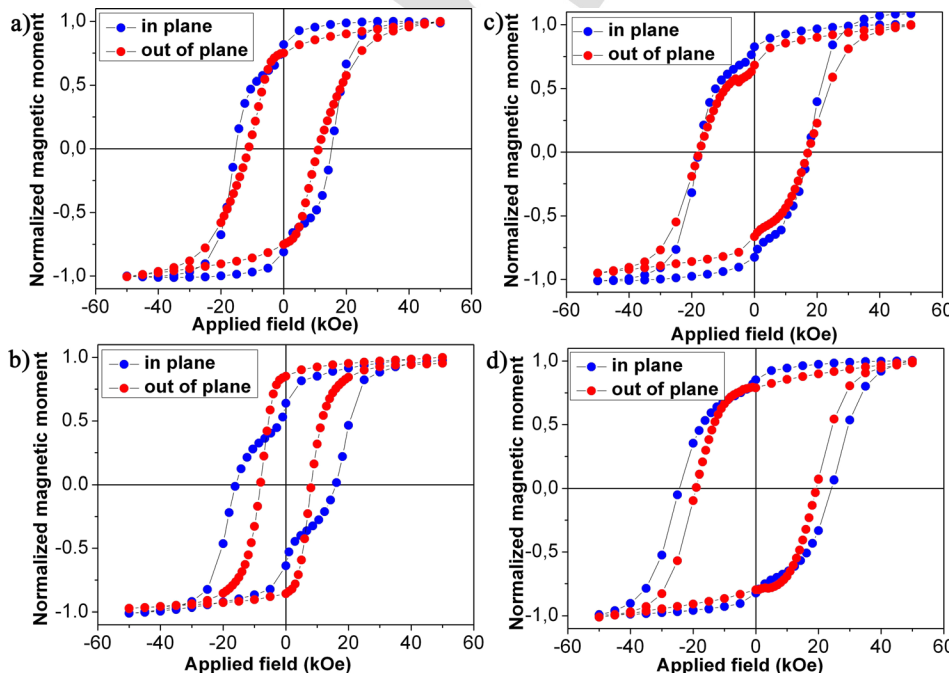


FIG. 8. SQUID-VSM M-H hysteresis loops of films annealed at different temperatures: (a) Pt/Fe at 973 K; (b) Pt/Fe at 1073 K; (c) Pt/Ag/Fe at 973 K; and (d) Pt/Ag/Fe at 1073 K.

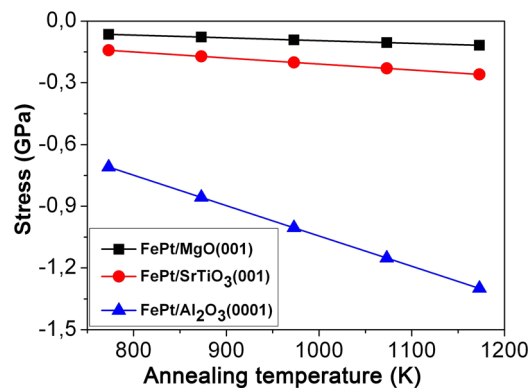


FIG. 9. Calculated thermal stresses arising from differences in thermal expansion coefficient between the substrate and the FePt film.

239 These stresses promote the chemical ordering and texture
240 formation in the films.

241 IV. CONCLUSION

242 In conclusion, sputtering of the Pt/Fe and Pt/Ag/Fe films
243 onto Al₂O₃(0001) single crystalline substrates and following
244 post-annealing leads to the formation of pronounced (001)-
245 texture. Furthermore, the onset temperature for chemical
246 ordering in these films is lower compared to films prepared
247 on MgO(001) and SrTiO₃(001) substrates. It is important to
248 note that there is no noticeable effect of the substrate choice
249 on the diffusion process but its influence on the chemical
250 ordering and texture formation is significant. Differences in
251 the structural properties of the films deposited onto different
252 substrates can be explained by the stress state that occurs
253 during post-annealing. Also it was shown that introduction
254 of the Ag intermediate layer is an effective method to
255 increase the coercivity of the film.

256 ACKNOWLEDGMENTS

257
258 This work was financially supported by the Deutsche
259 Akademischer Austauschdienst (DAAD) in the frame of the
260 Leonard-Euler-Program (Grant No. 50744282) and by the
261 Ministry of Education and Science of Ukraine (Grant No.
262 0212U008249). The research was also supported by the
263 TÁMOP-4.2.2.A-11/1/KONV-2012-0036 project, implemented
264 through the New Hungary Development Plan co-financed by
265 the European Social Fund, and the European Regional
266 Development Fund. The authors gratefully acknowledge the
267 support of the Hungarian Scientific Research Fund (OTKA)
268 through Grant No. NF 101329. This research was also
269 supported by the European Union and the State of Hungary,
270 co-financed by the European Social Fund in the framework of
271 the TÁMOP 4.2.4. A/2-11-1-2012-0001 “National Excellence
272 Program” (author G. L. Katona).

273
274 ¹D. Weller, O. Mosendz, G. Parker, S. Pisana, and T. S. Santos, “L₁₀
275 FePtX–Y media for heat-assisted magnetic recording,” *Phys. Status Solidi A*
276 **210**, 1245 (2013).

277 ²*Development in Data Storage, Materials Perspective*, edited by S. N.
278 Piramanayagam and T. C. Chong (John Wiley & Sons, Inc., New York,
279 2012).

- ³A. T. McCallum, P. Krone, F. Springer, C. Brombacher, M. Albrecht, E. Dobisz, M. Grobis, D. Weller, and O. Hellwig, “L₁₀ FePt based exchange coupled composite bit patterned films,” *Appl. Phys. Lett.* **98**, 242503 (2011). 280
281
282
⁴L. Zhang, Y. K. Takahashi, A. Perumal, and K. Hono, “L₁₀ FePt ordered high coercivity (FePt)Ag-C granular thin films for perpendicular recording,” *J. Magn. Magn. Mater.* **322**, 2658–2664 (2010). 284
285
286
⁵T. Bulat and D. Goll, “Large-area hard magnetic L₁₀-FePt nanopatterns by nanoimprint lithography,” *Nanotechnology* **22**, 315301 (2011). 287
288
⁶J. Lyubina, B. Rellinghaus, O. Gutfleisch, and M. Albrecht, in *Handbook of Magnetic Materials*, edited by K. H. J. Buschow (Elsevier, Amsterdam, 2011), Vol. 19, pp. 291–395. 289
290
⁷Y. K. Takahashi and K. Hono, “On low-temperature ordering of FePt films,” *Scr. Mater.* **53**, 403 (2005). 292
293
⁸P. Rasmussen, X. Rui, and J. E. Shield, “Texture formation in FePt thin films via thermal stress management,” *Appl. Phys. Lett.* **86**, 191915 (2005). 294
295
⁹M. L. Yan, N. Powers, and D. J. Sellmyer, “Highly oriented nonepitaxially grown L₁₀ FePt films,” *J. Appl. Phys.* **93**, 8292 (2003). 297
298
¹⁰Y. C. Wu, L. W. Wang, and C. H. Lai, “Low-temperature ordering of (001) granular FePt films by inserting ultrathin SiO₂ layers,” *Appl. Phys. Lett.* **91**, 072502 (2007). 299
300
¹¹Y. S. Yu, H.-B. Li, W. L. Li, M. Liu, and W. D. Fei, “Low-temperature ordering of L₁₀ FePt phase in FePt thin film with AgCu underlayer,” *J. Magn. Magn. Mater.* **320**, L125 (2008). 302
303
¹²C. Feng, Q. Zhan, B. Li, J. Teng, M. Li, Y. Jiang, and G. Yu, “Magnetic properties and microstructure of FePt/Au multilayers with high perpendicular magnetocrystalline anisotropy,” *Appl. Phys. Lett.* **93**, 152513 (2008). 304
305
¹³Y. S. Yu, H.-B. Li, W. L. Li, M. Liu, and W. D. Fei, “Structure and magnetic properties of magnetron sputtered [(Fe/Pt/Fe)/Au]_n multilayer films,” *J. Magn. Magn. Mater.* **322**, 1770 (2010). 306
307
¹⁴W. Y. Zhang, H. Shima, F. Takano, H. Akinaga, X. Z. Yu, T. Hara, W. Z. Zhang, K. Kimoto, Y. Matsui, and S. Nimori, “Enhancement in ordering of Fe₅₀Pt₅₀ film caused by Cr and Cu additives,” *J. Appl. Phys.* **106**, 033907 (2009). 308
309
¹⁵C. Feng, B.-H. Li, Y. Liu, J. Teng, M.-H. Li, Y. Jiang, and G.-H. Ya, “Improvement of magnetic property of L₁₀-FePt film by FePt/Au multilayer structure,” *J. Appl. Phys.* **103**, 023916 (2008). 310
311
¹⁶O. P. Pavlova, T. I. Verbitska, I. A. Vladymyrskyi, S. I. Sidorenko, G. L. Katona, D. L. Beke, G. Beddies, M. Albrecht, and I. M. Makogon, “Structural and magnetic properties of annealed FePt/Ag/FePt thin films,” *Appl. Surf. Sci.* **266**, 100 (2013). 312
313
¹⁷R. F. C. Farrow, D. Weller, R. F. Marks, M. F. Toney, A. Cebollada, and G. R. Harp, “Control of the axis of chemical ordering and magnetic anisotropy in epitaxial FePt films,” *J. Appl. Phys.* **79**, 5967 (1996). 314
315
¹⁸Y. K. Takahashi and K. Hono, “Interfacial disorder in the L₁₀ FePt particles capped with amorphous Al₂O₃,” *Appl. Phys. Lett.* **84**, 383 (2004). 316
317
¹⁹T. Bulat and D. Goll, “Temperature dependence of the magnetic properties of L₁₀-FePt nanostructures and films,” *J. Appl. Phys.* **108**, 113910 (2010). 318
319
²⁰Y. F. Ding, J. S. Chen, and E. Liu, “Epitaxial L₁₀ FePt films on SrTiO₃(100) by sputtering,” *J. Cryst. Growth* **276**, 111 (2005). 320
321
²¹M. Albrecht and C. Brombacher, “Rapid thermal annealing of FePt thin films,” *Phys. Status Solidi A* **210**, 1272 (2013). 322
323
²²L.-W. Wang, W.-C. Shih, Y. C. Wu, and C.-H. Lai, “Promotion of [001]-oriented L₁₀-FePt by rapid thermal annealing with light absorption layer,” *Appl. Phys. Lett.* **101**, 252403 (2012). 324
325
²³H. Oechsner, R. Getto, and M. Kopnarski, “Quantitative characterization of solid state phases by secondary neutral mass spectrometry,” *J. Appl. Phys.* **105**, 063523 (2009). 326
327
²⁴L. Péter, G. L. Katona, Z. Berényi, K. Vad, G. A. Langer, E. Tóth-Kádár, J. Pádár, L. Pogány, and I. Bakonyi, “Electrodeposition of Ni–Co–Cu/Cu multilayers: 2. Calculations of the element distribution and experimental depth profile analysis,” *Electrochim. Acta* **53**, 837 (2007). 328
329
²⁵H. Oechsner, R. Getto, and M. Kopnarski, “Quantitative characterization of solid state phases by secondary neutral mass spectrometry,” *J. Appl. Phys.* **105**, 063523 (2009). 330
331
²⁶I. A. Vladymyrskyi, O. P. Pavlova, T. I. Verbitska, S. I. Sidorenko, G. L. Katona, D. L. Beke, and I. M. Makogon, “Influence of intermediate Ag layer on the structure and magnetic properties of Pt/Ag/Fe thin films,” *Vacuum* **101**, 33 (2014). 332
333
²⁷D. Burgreen, *Elements of Thermal Stress Analysis* (C.P. Press, New York, 1971). 334
335
336
337
338
339
340
341
342
343
344
345
346
347
348
349
350
351