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The Use of Sodium to Calibrate the Transport Modeling of Water Pollution in Sandy Formations Around an **Uninsulated Sewage Disposal Site**

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15Abstract In the present paper we suggest a novel calibration method of the model for hydrodynamic and 16contaminant transport using the example of a sewage 17 18 disposal site set up uninsulated in a sandy environment. With the hydrodynamic model we applied time-19dependent model calculations in order to fit the individ-2021 ual hydrodynamic parameters. For the calibration of the transport model, sodium was chosen, which has a neg-22ligible retardation factor. We demonstrated that this 2324approach is suitable for creating a model that provides calculated results comparable to the actually measured. 25experimental ones. The created model proved to be

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appropriate for use in the estimation of the maximal 27spatial extension of the contamination, which-in the 28case of the investigated sewage disposal site-was 29found to be 0.1 km^2 in the near-surface (1–3 m deep) 30 layers, whereas it was three times higher at a depth of 3140–60 m. 32

Keywords Contamination transport · Groundwater 33 pollution · Model calibration · Sewage disposal site · 34 Sodium 35

1 Introduction

The anthropogenic contamination of groundwater pre-37 sents a serious problem in many locations worldwide 38(Datta et al. 2011; Delkash et al. 2014). Contaminants of 39agricultural and industrial origin can frequently be 40 found lying behind this problem (Al-Khashman 2008); 41 in addition, the lack of any sewerage installation, poorly 42insulated or even uninsulated communal septic tanks, 43 sewage disposal sites lacking proper technological pro-44 tection, and landfills for solid waste materials can also 45pose a threat and may seriously contaminate the ground-46 water (Phan et al. 2013, 2014). The rate and severity of 47 contamination can reach such levels that in and around 48settlements the groundwater may become unsuitable not 49only for human or animal consumption but also for 50irrigation (Somlyódi 2002). Of the various groundwater 51types, shallow groundwater is the most at risk, because it 52is usually located near the surface and contaminants 53from the topsoil are first transferred here via water 54

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infiltration (Cho et al. 2000; Kerényi 2003; Slack et al. 552005, 2007; Fejes et al. 2012). 56

57To control the quality of groundwater and to follow the transport of possible contaminants, the operation of a 58monitoring network is essential. At the same time, the 5960 interpretation of the measurement results requires the formulation of a conceptual geological model. If a math-61ematical model based on this conceptual model is also 6263 available, it enables significant further qualitative analyses. Over the last few decades, due to the rapid devel-64 opment of computer science, the creation of such 65 models-and their practical application in the solution 66 of groundwater-protection problems-has become more 67 and more frequent (Shahid et al. 2000; Sonneveld et al. 68 2010; Zhang et al. 2013). In order to protect groundwa-69 ter, it is vital to ensure both the geological and the 70technological protection of the waste disposal sites 71(Panno et al. 2002, 2006; Sipos et al. 2012; Chawla 7273and Singh 2014). This statement is supported by the research conducted by Regadío et al. (2011) and Zhan 7475et al. (2013) in sampling locations in Spain and southeastern China, respectively, in the areas surrounding 76operational waste disposal sites. In India, the high arse-7778 nic (As) content of the groundwater can cause serious 79problems, and to understand these processes, researchers have relied on three-dimensional hydrody-80 namic and transport models that work with finite-81 difference methods, which have long been established 82in Japan (Nakaya et al. 2011). Chakraborty and Ghosh 83 (2009) also worked with software that uses a finite-84 difference method, to model chloride and sodium trans-85port in the vicinity of a municipal solid waste landfill. 86

The present research was conducted in the area of a 87 sewage disposal site that had been operational for almost 88 three decades and then recultivated. Based on the avail-89 able measurement results, the main aim of our study was 90 to establish a calibration procedure that would signifi-91cantly improve the reliability of the contaminant trans-9293 port model currently in use. A further aim was to use this newly calibrated model to assess the main characteris-94tics of the spatial and temporal changes in water quality 95and the development of the clarifying process following 96 recultivation. The latter is even more important as there 97 is surprisingly little information available in the litera-98 ture with regard to the actual size of the affected area of 99 such a contamination source in sandy zones. The rate of 100 101 water quality improvement after a recultivating inter-102vention is also rather unclear. Thus, these investigations may provide a solid basis for the assessment of risks due 103

to the contamination of artesian waters and also for the 104establishment of the need for a possible future 105intervention. 106

2 Materials and Methods 107

2.1 Description of the Study Area 108

The area we studied is found in eastern Hungary, 8 km 109south of Debrecen, in the administrative township of 110Mikepércs (Fig. 1). Regarding its geological structure, it 111 is a S-SW sloping, sand-covered alluvial fan plain of 112aeolian origin. The climate of the area is warm temper-113ate and dry temperate; the mean annual temperature is 114 9.8-10.0 °C, and the average annual rainfall is 550-115580 mm (Dövényi 2010). Between 1982 and 2011, the 116hole of an abandoned sandpit was used as a disposal site 117for the wastewater produced by the town, the quantity of 118which even exceeded a daily 100 m³. This sewage 119disposal site had been established and used without 120 any authorization, and had no technological protection. 121The sewage disposal site with a total capacity of 12220,000 m³ was formed on a sandy soil structure, and 123under such conditions the contaminants could easily 124infiltrate into the groundwater. Previous studies from 1252004, 2005, and 2010 justified that, as a consequence 126of the sewage disposal, the groundwater became heavily 127contaminated (Szabó et al. 2007a; Szabó et al. 2007a; 128Szabó et al. 2007b). The recultivation of the sewage 129disposal site took place in the spring of 2011. 130According to the recultivation plan, the wastewater from 131the sewage disposal site was transferred to a sewage-132treatment plant operated by the Debreceni Vízmű Zrt. 133(Debrecen Waterworks Ltd.). The sewage sludge was 134processed and rendered harmless by the AKSD 135Városgazdálkodási Kft. (A.K.S.D. City Management 136Ltd. of Debrecen). Based on the recultivation plans, 137the pit of the former sandpit-after being cleaned of 138 wastewater and sewage sludge-was filled with 9000-1399500 m³ of sand and 2500 m³ of organic matter rich soil 140 (Green Side 2008). 141 Q3

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2.2 Sampling and Laboratory Tests

In order to collect soil and groundwater samples, a total 143of 12 shallow groundwater wells have been drilled since 144 2004 with an Eijkelkamp-type hand auger. Taking the 145NE-SW flow direction of the groundwater into 146



Fig. 1 The location of the study area, the sampling points, and the vertical sections (A-B, C-D)

147consideration, the sampling locations (see Fig. 1) were 148assigned so that the samples would provide as much information as possible about the groundwater condi-149tions in the surroundings of the sewage disposal site. 150The well casing was made of PVC pipes with a diameter 151of 50 mm, the lower 1 m section of which is screened. 152This was necessary to prevent the boreholes becoming 153silted up when groundwater flows into the well. The 154bottom depth of the wells was 1 m below the ground-155water level measured at the time of the installation 156157(Table 1).

Soil samples were taken from each borehole at every 20 cm, and were then sealed in plastic bags and transferred to the laboratory of the Department of Geology of the University of Debrecen. The samples were exsiccated at 60 °C, and then their granulometric composition was determined by the Köhn-pipette method (Müller et al. 2009).

For the water sampling a peristaltic pump was used. 165Before sampling we extracted three times the volume of 166water originally contained in the well to obtain water 167from the pore volume. At each sampling time one water 168sample (250 ml) was taken from each well and the depth 169170of the groundwater level was measured (Table 1). The bubble-free water samples were transferred in tightly 171closed plastic bottles to the Geology Laboratory of the 172

University of Debrecen, where the Na concentration 173 was measured by using a PerkinElmer 3110 AAS. 174

The results of measurements were recorded in a175Microsoft Excel database. The statistical analyses were176done using SPSS 20. The maps were edited with the177Surfer software (version 8.0) (Golden Software LLC),178by Kriging interpolation.179

2.3 Model Calculation 180

After the recultivation, we continued the investigation of 181 the level of groundwater contamination. Based on the 182test results from three consecutive years after the 183recultivation, no relevant changes in the groundwater 184quality were found. We assume that in the course of the 185three-decade operation of the sewage disposal site both 186the sediment layers and the groundwater itself became 187 contaminated to such an extent that the long-term effect 188of the operation must also be taken into consideration. 189

In order to interpret the measurement results and to 190 predict the longer-range transport of the contaminants 191 via the groundwater, we performed hydrodynamic and 192 contaminant transport model calculations. The horizon-193 tal dimension of the modeled area was $1.8 \text{ km} \times 1.4 \text{ km}$ 194 (2.52 km²), whereas its vertical extension reached 115 m 195 above the level of the Baltic Sea. The HD72/EOV 196

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t1.1 t1.2	Table 1 The basic properties ofthe monitoring wells and main	ID	EOV Y	OV Y EOV X	Bottom depth (cm)	Depth of groundwater (cm)			
t1.3	statistical parameters of the depth of groundwater between 2004 and 2014					Mean	SD	Min.	Max.
t1.4	2017	MPL1	846942	236792	420	149.8	45.3	57	267
t1.5		MPL2	847171	236824	370	205.4	10.8	186	224
t1.6		MPL3	846816	237084	320	238.2	26.1	167	288
t1.7		MPL4	847191	237425	420	288.1	36.4	171	350
t1.8		MPL6	846837	236745	420	326.8	24.6	273	366
t1.9		MPL8	846706	236703	340	256.3	22.7	191	302
t1.10		MPL12	846758	236518	570	384.3	59.3	301	451
t1.11		MPL13	846904	236682	400	284.8	40.6	213	360
t1.12		MPL14	846593	236312	300	193.1	46.2	99	260
t1.13		F1	846892	236763	534	216.6	44.6	91	269
t1.14		F2	846874	236844	552	229.7	57.1	131	415
t1.15 Q4	EOV Egységes Országos Vetület	F3	846921	236889	531	339.5	37.1	253	400

coordinate system (EPSG 23700) was applied for the 197 198spatial coverages, in which one unit equals 1 m. In order 199to construct the digital relief map model of the surface, we used the results of our on-site measurements per-200 formed by two Trimble S9 dual-frequency, high-201 precision geodesic GPS instruments. The accuracy of 202the GPS measurements was 2 cm. The interpolation of 203the surface was completed with a free triangular mesh. 204

Based on the analysis of the boring log (Marton 2052009) of the confined aquifer well with a bottom depth 206of 258.6 m in the vicinity of the sewage disposal site, 207five different geological strata were identified in the 208209 model. In the near-surface layers, the dominant fraction of the geological structure is sand; however, 54 m below 210the surface, 50-m-thick clayey and silty strata appear 211212(Szabó and Szabó 2005; Marton 2009). The determina-213tion of the model's hydrodynamic factors-the diagonal elements of the hydraulic conductivity tensor (K_{xx} , 214 $K_{yy} \sim 10^{-5} - 10^{-7}$ m/s, $K_{zz} \sim 10^{-5} - 10^{-8}$ m/s) and the spe-215cific storage factor ($S_{\rm s} \sim 10^{-5} \text{ m}^{-1}$)—were based on the 216particle size distribution of the samples taken from the 217self-produced shallow groundwater wells, as well as on 218the data from the confined-aquifer well. For the hydrau-219lic conductivity determination, we have used the 220Campbell formula: 221

$$K = 4 \times 10^{-5} \times e^{-6.9 f_{\text{clay}} - 3.7 f_{\text{silt}}}$$

where f_{clay} and f_{silt} are the clay and silt fractions, respectively (Campbell 1985). 224

Figure 2 shows the grain size distributions along the 225 section of three selected monitoring wells (MPL1, 226 MPL2, MPL3). All the other wells show similar distribution. The graphs show that the dominant fraction is 228



Fig. 2 The grain size distributions of the MPL1, MPL2, and MPL3 wells

fine sand (over 80 %) and the fraction of coarse sand
falls below 20 %. The sum of the fractions of silt and
clay, which essentially determines the hydraulic conductivity, remains below 20 % in each layer of each well.

In order to calibrate the system's hydrodynamic pa-233rameters, we first made temporally varying hydrody-234namic model calculations. This was possible because 235the required set of rainfall and evapotranspiration input 236data for 3 years after the recultivation of the disposal site 237was available, as were the results of the hydrodynamic 238piezometry measurements necessary for the calibration 239and validation of the model. During the process of 240241modeling, we solved Eq. (1):

$$\frac{\partial}{\partial x}\left(K_{xx}\frac{\partial h}{\partial x}\right) + \frac{\partial}{\partial y}\left(K_{yy}\frac{\partial h}{\partial y}\right) + \frac{\partial}{\partial z}\left(K_{zz}\frac{\partial h}{\partial z}\right) - w = S_s\frac{\partial h}{\partial t}(1)$$

where w is the volumetric flux per unit volume 06 243 representing the sources and sinks of water and h is 244 the hydraulic head. During the transient calculations, 245 246 the hydrodynamic potential values at the circumference 247 of the sampling area were set so that they would approximate the water levels measured in the monitoring 248 249 wells closest to the edge of the area (ID: MPL-4 and 250 MPL-14) as much as possible. The hydrodynamic model was run by the USGS MODFLOW 2005 numerical 251 engine (Harbaugh 2005). Figure 3 shows the input 252 253 infiltration and evapotranspiration data of the timedependent model calculations, as well as the calculated 254

results of the hydrodynamic potential, in comparison 255 with the measured data from a few selected monitoring 256 wells. 257

In the course of the calibration, three of the studied 258model parameters-the infiltrated percentage of total 259rainfall, the average rate of evapotranspiration, and the 260hydraulic conductivity of the second stratum-were set 261in such a way that in the other monitoring wells the sum 262of root-mean-squares (RMS) of the measured and cal-263culated water levels-averaged to all sampling times-264were as low as possible. In Fig. 4 a calibration curve 265relating to a sampling time (day 446), which is selected 266 as an example, shows that the normalized RMS value is 267quite low (6 %) and the correlation coefficient approx-268imates 1 (r=0.966). Thus, the model can be considered 269reliable. 270

We created the steady-state hydrodynamic model271using the optimized parameters of the transient model272and the average annual infiltration typical of the area.273The solution of the steady-state model was used in the274contaminant transport calculations.275

For the contaminant transport calculations, sodium 276was chosen because of its low retardation factor 277 $(R \sim 1.5)$ (Rowe et al. 2004); i.e., it is relatively conser-278vative, so it does not transmute during transport via 279groundwater, and it is poorly adsorbed (Greenwood 280 and Earnshaw 1999; Sayyed and Bhosle 2011). 281Another important condition was that the given contam-282 inant should originate primarily from the wastewater 283





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Fig. 4 One calibration curve (relating to day 446 of sampling) of the transient hydrodynamic model







collected from the disposal site. When modeling thecontaminant transport, we used Eq. (2):

$$\frac{\partial(\theta C)}{\partial t} = \frac{\partial}{\partial x_i} \left(\theta D_{ij} \frac{\partial C}{\partial x_j} \right) - \frac{\partial}{\partial x_i} (\theta v_i C) \tag{2}$$

where *C* is the dissolved concentration of sodium; θ is the porosity of the subsurface medium; *t* is time; x_i is the distance along the respective Cartesian coordinate axis; D_{ij} is the hydrodynamic dispersion coefficient tensor; and v_i is the Darcy velocity, which can be determined from the



solution of Eq. (1). We determined the diffusion constant292that influences the contaminant transport as the molecular293diffusion coefficient of sodium in water: 4×10^{-9} m²/s.294The longitudinal, horizontal, and vertical dispersivities295were specified as 10, 1, and 0.1 m, respectively.296

To calculate the contaminant transport, the 297 MT3DMS v. 5.2 numerical transport engine (Zheng 298 and Wang 1999) was applied. In the course of the model 299 calibration, the parameters to be fitted were the Na 300 concentrations in the northern and southern volumes of 301 the disposal site. The transport calculations (Fig. 5) 302



Max. Residual: -82.886 (mg/L) at MPL-13/A Min. Residual: -1.156 (mg/L) at MPL-6/A Residual Mean: -18.632 (mg/L) Abs. Residual Mean: 31.134 (mg/L) Num. of Data Points : 12 Standard Error of the Estimate : 10.687 (mg/L) Root Mean Squared : 40.043 (mg/L) Normalized RMS : 10.013 (%) Correlation Coefficient : 0.965 303 were calibrated to the sampling time following 304 recultivation, for which reliable measurement data 305 were available. Based on the graph it can be 306 established that the reliability of the transport model 307 is acceptable, since the normalized RMS is around 308 10 % and the value of the correlation coefficient 309 (r=0.965) approximates to 1.

310 **3 Results and Discussion**

311In the first stage we had to verify that sodium is also 312suitable for the calibration of the contaminant transport model in the sense that it originates primarily 313from the wastewater deposited at the disposal site 314(Szabó and Bessenyei 2013). This is because in al-315kaline soils the sodium contents of the groundwater 316may increase under natural conditions as well. 317 However, based on the test results of the water sam-318ples originating from the monitoring wells, it can be 319320 clearly established that in the affected area of the 321sewage deposit it is not necessary to take into account 322 any alkaline-formation process, since in the wells 323 located further away from the disposal site (MPL14, 324MPL2, MPL4) we consistently measured quite low (under 20 mg/l) sodium content. Values greater by 325 about one order of magnitude (150-500 mg/l) were 326 327 found without exception in the wells in the immediate vicinity of the sewage disposal site (MPL6, F2, 328 MPL1, F1, F3, MPL13) (see Figs. 1 and 6). 329

Regarding sodium transport, we used the model to
investigate two consecutive time periods. For the first
period we assumed a disposal site that had been continuously operational for three decades and filled with

contaminants. The second period refers to the disposal334site that had been recultivated and filled with sand after335the removal of the contaminants.336

In the 30-year operational period, the hydrodynamic 337 potential field was calculated for the static condition, 338 with the important difference in comparison to the transient model that in the area of the disposal site we 340 specified the water level that was available based on 341 the measurement data from the operational period (see Fig. 7a). 343

In the operational period, at the border of the 344 saturated and unsaturated zones, the maximal hori-345 zontal infiltration speed-according to calculations 346based on the model—was 4.2×10^{-6} m/s, whereas 347 the maximal vertical infiltration speed was lower, at 348 3.1×10^{-6} m/s in areas close to the sewage disposal 349site (Fig. 7a). In the period following recultivation, in 350both the horizontal $(1.3 \times 10^{-6} \text{ m/s})$ and the vertical 351 $(9.2 \times 10^{-7} \text{ m/s})$ infiltration speed, a certain decrease 352was detected, which can be explained by the cessa-353 tion of the dome-like hydrodynamic potential field 354that came about as a consequence of the sewage 355disposal (Fig. 7b). 356

When simulating the contaminant transport, in the 357 dome-like potential field typical of the operational 358period, in the northeastern 3/4 and southwestern 1/4 359volumes of the disposal site, constant sodium con-360 centrations of 200 and 500 mg/l were assumed, re-361spectively. The sewage disposal was divided into two 362 parts by an earth bank, and the two catch basin parts 363 became isolated from each other whenever the water 364levels were low; therefore, in the southwestern basin 365part increased evaporation could occur, which result-366 ed in an increase in the Na concentration (Fig. 8a). 367

Fig. 6 The sodium contents of the water from shallow groundwater wells, based on the measurements performed during the study period (2004–2014). The extreme values (third quartile [Q3]+3 interquartile range [IQR]) were not plotted in the graph







Fig. 7 a The W-E orientated vertical picture of the dome-like hydrodynamic potential field in the course of operation. **b** The W-E orientated vertical picture of the hydrodynamic potential field after recultivation

368 The wastewater had been disposed in the larger, 369 northeastern basin part. Consequently, the possibility

370 of a continuous thinning was present.

During the functional period, in the groundwater 371 dome that was being formed by the disposed wastewater, a radial transport of contamination can be observed; 373



Fig. 8 The horizontal transport of Na contamination, 3 m below the surface: **a** the beginning of operation (1982); **b** 1st year of operation (1983); **c** 10th year of operation (1992); **d** end of

operation, recultivation (2011); **e** 1st year after recultivation (2012); **f** 5th year after recultivation (2016); **g** 10th year after recultivation (2021); and **h** 30th year after recultivation (2041)

therefore, the NE-SW flow direction, which is charac-374teristic under natural conditions, was hindered. The 375height of the groundwater dome in the surrounding areas 376exceeded the average natural groundwater level by 1 m 377 on average. According to our calculations, the contam-378 379 ination transport at a depth of 3 m reached a distance of 50-100 m by the end of the first year of contamination 380 (Fig. 8b). At such distances the Na concentration de-381 creased to the level of the background concentration 382level (20 mg/l) typical of the area. After 30 years had 383 passed (by the end of the operational period), at such a 384depth below the surface the Na contamination-accord-385ing to both the model calculations and the test results of 386 the collected water samples-could be detected at a 387distance of up to 200 m, and the size of the affected 388 contamination area was 112,000 m² (Fig. 8d). Sodium is 389 a conservative, i.e., enduring, contaminant, and its rate 390 of release from the sewage disposal site is more or less 391392time-independent; therefore, the limited horizontal dimension of the contamination suggests that in the near-393 394surface layers the vertical component must be given 395 greater emphasis in the transport process.

In the calculations for the 30-year period after 396 397recultivation, the starting contaminant concentration 398 was borrowed from the concentration distribution results of the end of the first period, except for the volume 399 size of the disposal site itself, where we defined a clean, 400 401 contaminant-free territory filled with sand, since in the course of recultivation both the wastewater and the 402 sewage sludge were removed from the disposal site 403 and the cleaned pit was filled with pure sand. With these 404results we predicted the horizontal transport of the con-405 taminants for 30 years. In this period the natural, the NE-406 SW groundwater flow direction already prevails. Since 407 the resupply of contaminants has ceased, the model 408 409 predicts the gradual decrease in the Na concentration. Based on the model simulations, after a year a 350-410400 mg/l Na concentration can still be expected in the 411 412southwestern corner of the disposal site (Fig. 8e). This result is further supported by the 393.3 mg/l concentra-413414 tion value measured in the MP6 monitoring well in July 2012. According to the model, in the fifth year after 415recultivation the Na concentration would decrease to 416 250-300 mg/l (Fig. 8f), and in 10 years' time in the 417southwestern corner of the sewage disposal site the 418 model predicts a mere 140-160 mg/l concentration 419420 (Fig. 8g). The model indicates that after 30 years have passed, in the areas surrounding the above-mentioned 421 location, in a 26,000 m^2 area this value (20-40 mg/l) 422

will still exceed the limit values to a certain extent, but423apart from this, we can base our calculations on concen-424trations below 20 mg/l, even for the former sewage425disposal site itself. In other words, during the 30-year426period following recultivation, the size of the contami-427nated area would gradually decrease and also slowly428migrate in a SW direction (Fig. 8h).429

In addition to its horizontal transport, the vertical 430 distribution of the contamination was also modeled. 431Figure 9a-d demonstrates the outcome of the vertical 432transport of the contamination, in the 30-year time span 433 from the start of operation as a functional sewage dis-434 posal site to the beginning of its recultivation. It can be 435clearly seen that the contamination already approaches 436 the 54-m-deep water-impermeable layer in the first year 437(Fig. 9b). At such a depth the vertical transport of the 438contaminants slows down, and by the end of the oper-439ational period (i.e., in the 30th year following the start of 440 its functioning) it reaches down to about 70 m (Fig. 9d). 441 The highest Na concentrations can be detected under the 442 south basin part of the sewage disposal site, due to the 443 concentration by evaporation mentioned earlier. 444 Concentrations higher than 400 mg/l were undetectable 44540 m below the surface. It is also clearly observable that 446 because of the continuous contaminant resupply and the 447 water-impermeable layer found at a depth of 54 m, 448 about 50 m below the surface a more intensive horizon-449 tal spread has developed. In this zone, the polluted water 450can reach distances as great as 400 m, which signifi-451cantly exceeds those calculated for near-surface layers. 452The horizontal transport is more intensive toward the S-453SW, since this fits the local groundwater flow direction. 454

In the first year after recultivation, no significant 455changes were observed. However, as Fig. 9e clearly 456shows, the contamination rate significantly decreased 457in the near-surface layers; i.e., the 400-500 mg/l Na 458concentration was detected only in a much narrower 459strip. In the 5 years after recultivation, the Na concen-460 tration reduces to below 300 mg/l in the near-surface 461 layers even in the most heavily contaminated parts. 462 However, in the zone lying between 30 and 40 m below 463 the surface, below the southern basin of the sewage 464 disposal site, the model still indicates a value of 465 350 mg/l (Fig. 9f). After 10 years, the contamination 466 decreased further: the highest concentration was 467 290 mg/l, which appeared at greater depth, between 35 468 and 55 m below the surface (Fig. 9g). In the 30th year 469 following recultivation, some contamination can still be 470 detected, but its level never exceeds 210 mg/l. The most 471 #### Page 10 of 13



Fig. 9 The vertical transport of contamination: **a** the beginning of operation (1982); **b** 1st year of operation (1987); **c** 10th year of operation (1992); **d** end of operation, recultivation (2011); **e** 1st

heavily polluted zone shifted deeper, appearing in the
strip 50–60 m below the surface. However, in the nearsurface layers, these values fell to around or below the
background concentrations (Fig. 9h).

The migration of the contamination in a S-SW direction was considerable even in the period after
recultivation. The infiltration continued to be most intensive above the water-impermeable layer that runs
54 m below the surface. After 10 years, water with a

year after recultivation (2012); **f** 5th year after recultivation (2016); **g** 10th year after recultivation (2021); and **h** 30th year after recultivation (2041)

higher than 20 mg/l Na concentration reached as far as 481about 500 m (Fig. 9g). Unfortunately, based on this 482model, we could not establish precisely how far the 483contaminated water may travel in 30 years, because 484the size of the modeled area does not allow this. 485However, we can assume that the distance covered 486 may be somewhere around 600-700 m. Although the 487 dominant transport direction is S-SW, due to diffusion 488 the contamination can spread toward the north as well. 489



Fig. 10 The measured and the calculated values for the first year after the recultivation

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However, the rate of this process is a mere 7 m per
30 years. The water-impermeable layer considerably
hindered and slowed down the vertical transport of
contaminants, but during the 30 years following
recultivation, the contamination reached a further 6 m
down (i.e., 76 m below the surface).

In order to verify the reliability of our model, we 496 compared our measured results in the first year after the 497 recultivation to the calculated values of the model for 498the end of the first year after the recultivation. In Fig. 10 499it can be seen that the calculated values correspond with 500 the measured ones, even though some differences may 501502 occur because of the uncertainty of the sampling and 503laboratory measurements as well as the model 504calculation.

505 4 Conclusions

In the present study we suggested a novel calibration 506 507 method of the model for hydrodynamic and contami-508nant transport, shown via the example of a sewage disposal site set up uninsulated in a sandy environment. 509 510In the case of the hydrodynamic model, we applied 511time-dependent model calculations in order to fit the individual hydrodynamic parameters. For the calibra-512tion of the transport model, sodium-which has a neg-513514ligible retardation factor-was used.

We proved that the Na content of the groundwater-515which exceeded the 20 mg/l background concentration 516and close to the disposal site even reached 500 mg/l-517definitely originated from the sewage disposal site. 518Consequently, sodium-due to its low retardation fac-519tor-proved to be a suitable element to allow us to 520estimate and predict the maximal spatial extension of 521522the contamination. We established that by the end of the operational period the polluting effect of the sewage 523disposal site remained detectable within a 200-m range 524in the near-surface (1-3 m deep) layers, which is equiv-525alent to an area slightly greater than 0.1 km². However, 52652740-50 m below the surface, the contaminated area is substantially-at least three times-greater than that 528measured near the surface. After recultivation the resup-529ply of contamination ceased and according to the model, 530by the end of the modeled period the size of the polluted 531area would diminish to one-fifth in the near-surface 532533layers and the Na concentration would decrease approx-534imately to the level of the background concentration. 535However, in the deeper layers (40–50 m below surface)

the Na concentration would still remain higher by about 536 one order of magnitude. 537

In this pilot study we have shown that the reliability 538of contamination transport model for a sewage disposal 539site can be justified. The developed model calibration 540method can be profitably applied to similar relatively 541small-scale sites to estimate the maximal spatial exten-542sion of contamination in a sedimentary environment, 543and it may provide a proper basis for modeling the 544transport of polluting agents that are present in similar 545sewage disposal sites, which may pose a significantly 546 higher risk in terms of human health and safety. 547

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